

New Au(I) Complexes for Homogeneous Enantioselective Catalysis



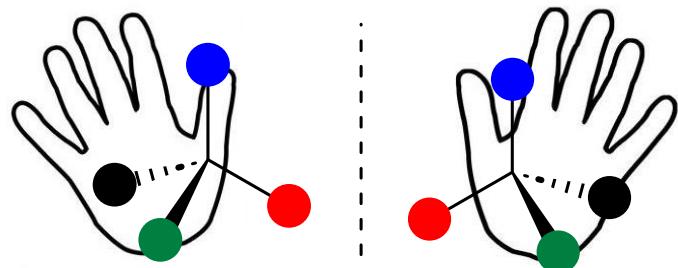
Xavier Guinchard

ICSN, Gif sur Yvette

Phosphorus Chemistry and Catalysis group

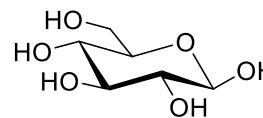
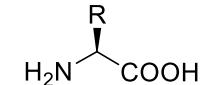
Chirality in nature

1) Organic molecules are chiral



2 enantiomers are mirror images

2) Nature is asymmetric

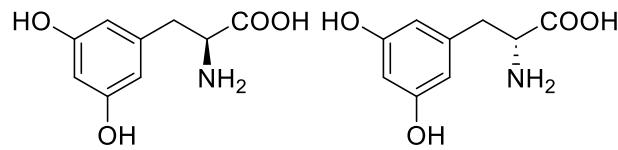


Aminoacids are all (*S*)-configured

Sugars are all *D*

Proteins, nucleic acids are all chiral molecules

3) Configuration matters for bioactivity

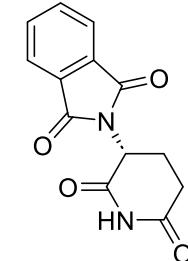


(*S*)-dopamine

Médicament
Parkinson

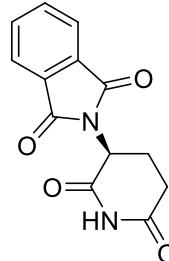
vs

Poison



(*R*)-thalidomide

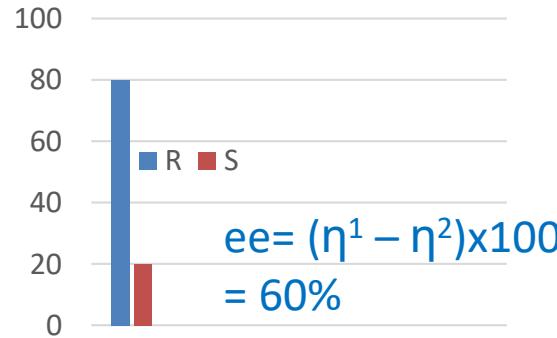
Médicament
sédatif



vs

Tératogène

4) Enantiomeric excess

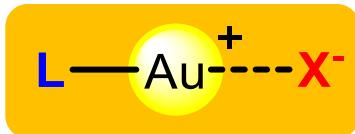


It is crucial to be able to access both enantiomers of a given molecule

The highest the ee,
the better!

Au(I) catalysis

Structure

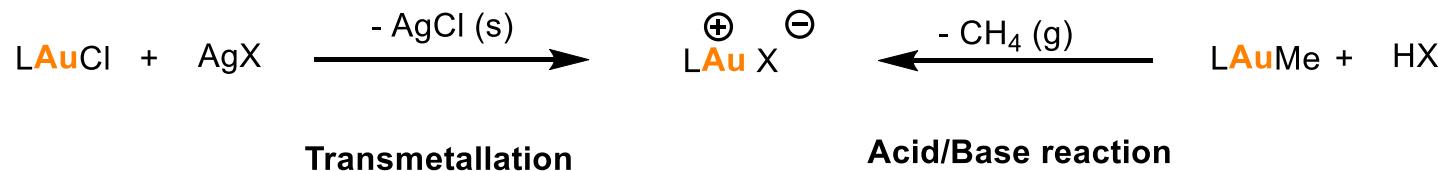


L : Phosphorus ligand or NHC

X : Weakly coordinating counterion
(Cl >> BF₄, PF₆, SbF₆, NTf₂...)

→ Complexes for homogenous catalysis!

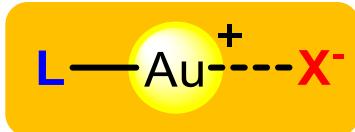
Activation of the complexes



→ It is necessary to activate the complex for catalytic purposes

Au(I) catalysis

Structure



L : Phosphorus ligand or NHC

X : Weakly coordinating counterion
(Cl >> BF₄, PF₆, SbF₆, NTf₂...)

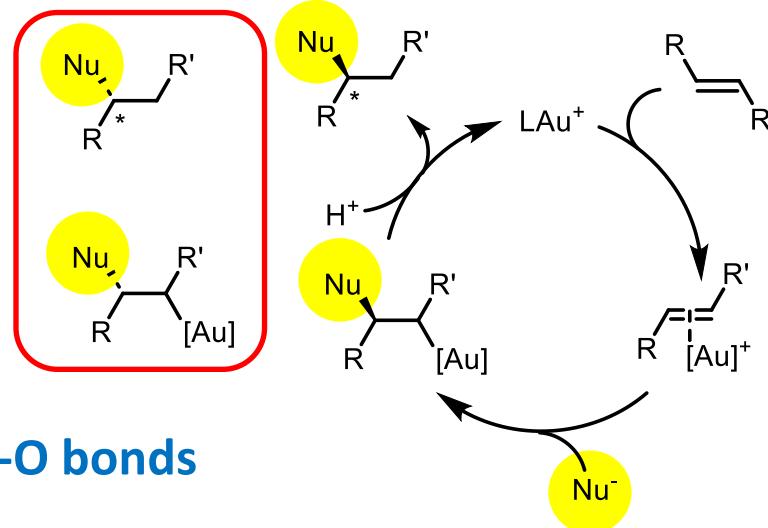
→ Complexes for homogenous catalysis!

Reactivity

Main property: Carbophilic Lewis acids

Main applications: To add nucleophiles to unsaturated bonds

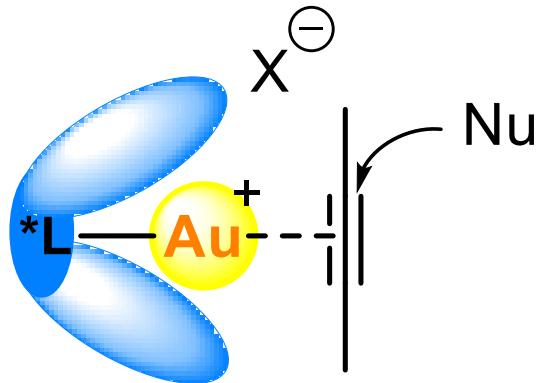
How can we favor the formation of only one enantiomer?



→ **Creation of new C-C, C-N, C-O bonds**

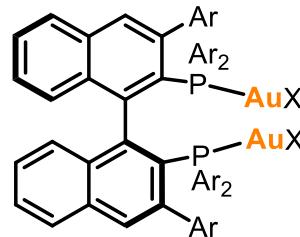
Strategies in enantioselective Au(I) catalysis

Chiral ligand-based approach

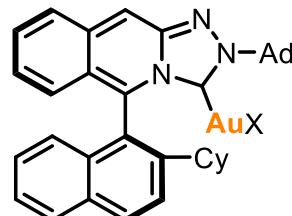


- The chiral ligand creates an asymmetric environment
- The linearity of Au is a limitation
- Bulky ligands are required

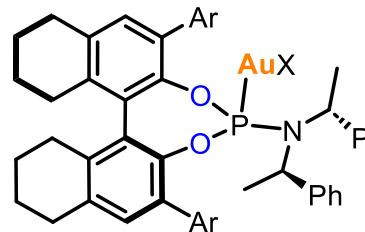
Representative chiral complexes



di-phosphine



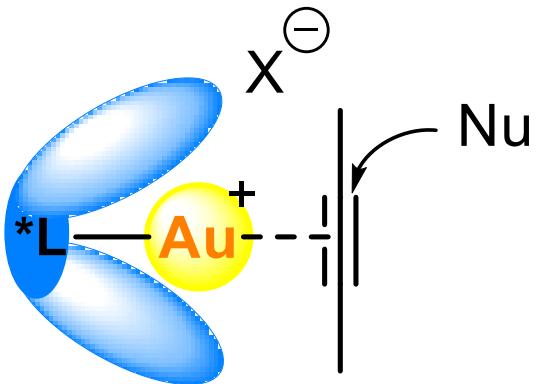
NHC



phosphoramidites

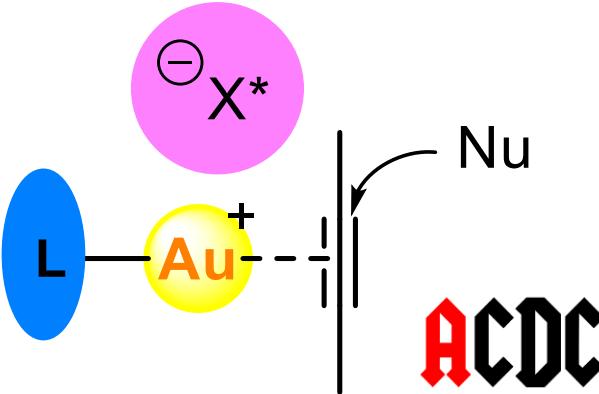
Strategies in enantioselective Au(I) catalysis

Chiral ligand-based approach



- The chiral ligand creates an asymmetric environment
- The linearity of Au is a limitation
- Bulky ligands are required

Chiral Counterion-based approach

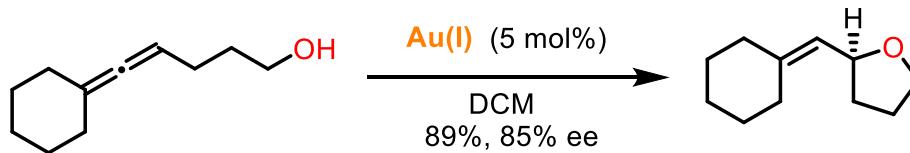
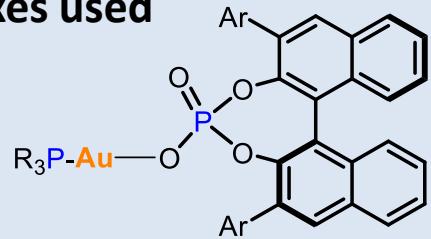


- The chiral counterion is closer to the reacting center
- Asymmetric Counterion-Directed Catalysis strategy (List/Toste)

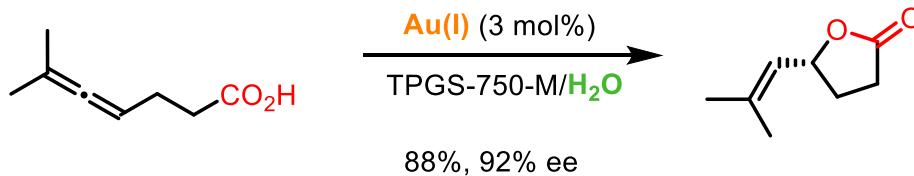
General reviews: Toste, *Chem. Soc. Rev.* **2016**, 45, 4567. b) A. Pradal, P. Y. Toullec, V. Michelet, *Synthesis*, **2011**, 1501. c) Patil, N. T. et al. *Isr. J. Chem.* **2023**, 63, e202200039.

Chiral Au(I) phosphates

Chiral complexes used



Reactivity

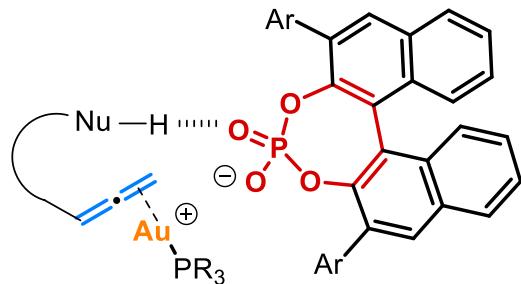


Hamilton, G.L., Kang, E.J. Mba, M. Toste, F.D. *Science* **2007**, *317*, 496.
S. Handa, D. J. Lippincott, D. H. Aue, B. H. Lipshutz, *Angew. Chem. Int. Ed.* **2014**, *53*, 10658.

Chiral Au(I) phosphates

Plausible scenario for intramolecular hydrofunctionnalizations

ACDC

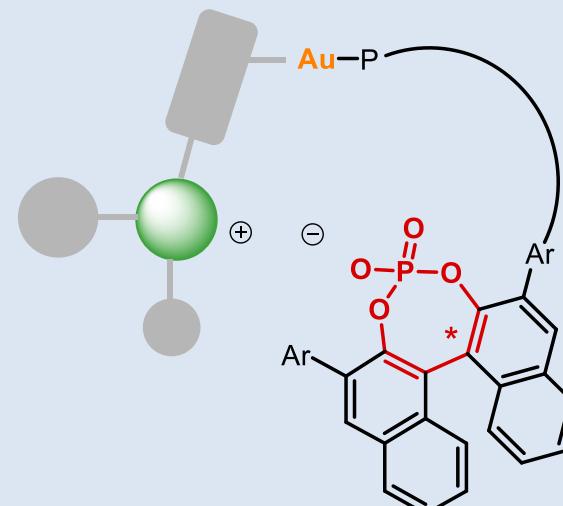
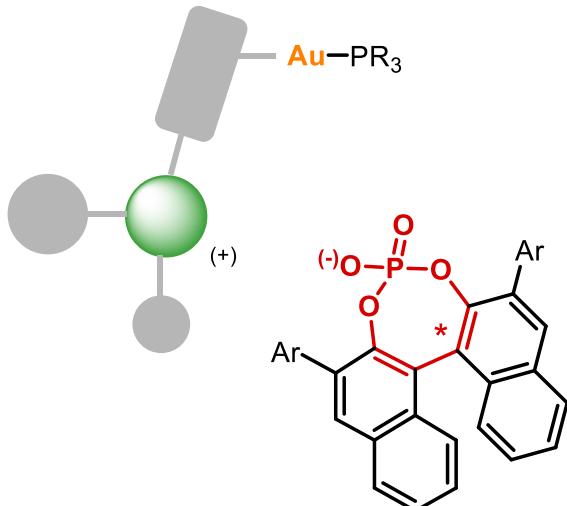


**Tethered Counterion
Directed Strategy (TCDC)**

~~A~~ CDC

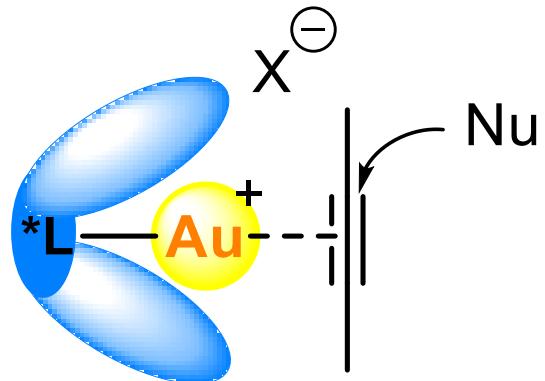
- Enhance rigidity
- Geometrical constraints
- Molecular organization

Other reactions involving ion pairs

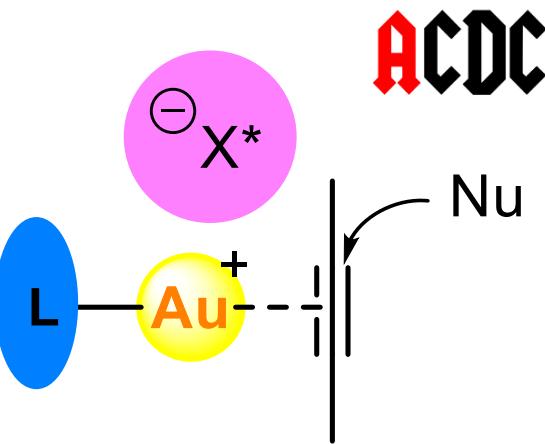


Strategies in enantioselective Au(I) catalysis

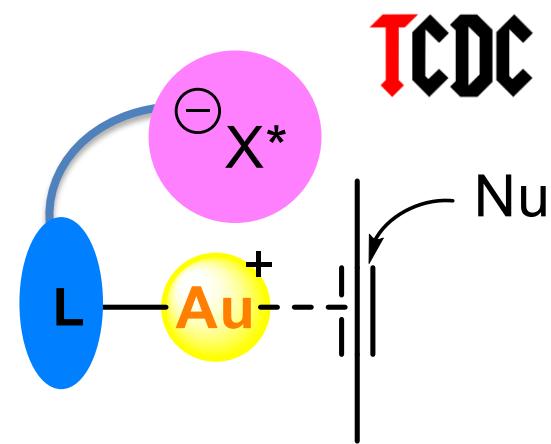
Chiral ligand-based approach



Chiral Counterion approach



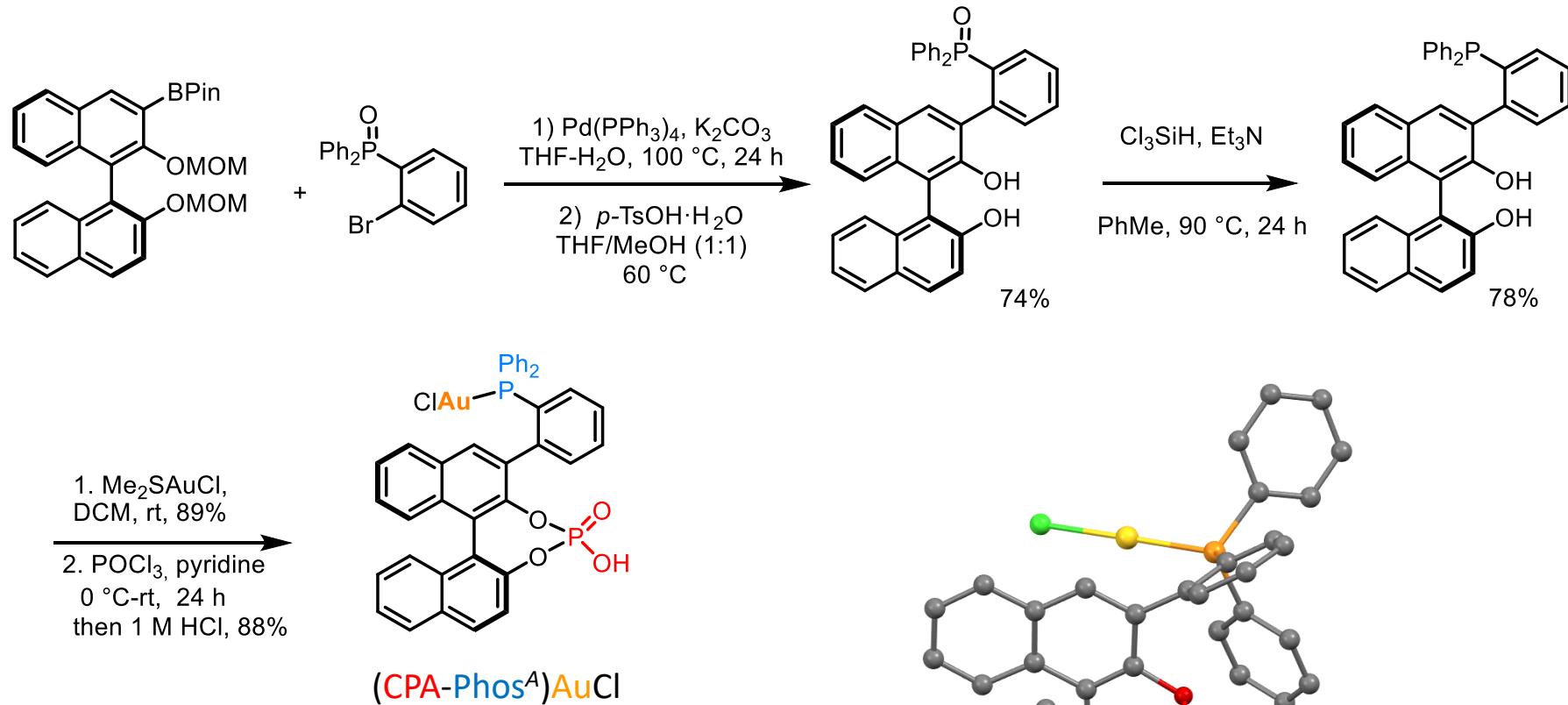
Tethered approach



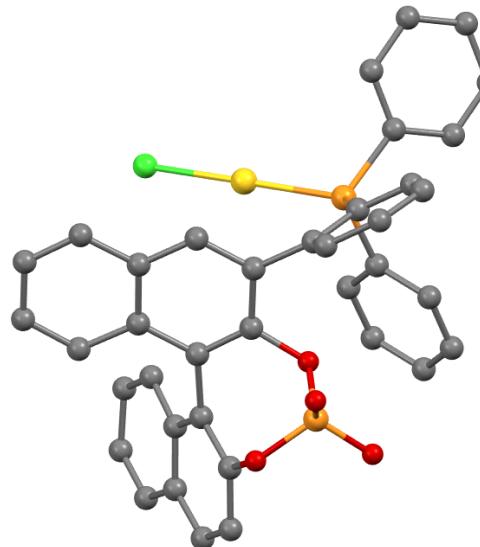
General reviews: Toste, *Chem. Soc. Rev.* **2016**, 45, 4567. b) A. Pradal, P. Y. Toulléc, V. Michelet, *Synthesis*, **2011**, 1501. c) Patil, N. T. et al. *Isr. J. Chem.* **2023**, 63, e202200039.

Synthesis of the chiral Au(I) precatalyst

Zhenhao Zhang
Vitalii Smal

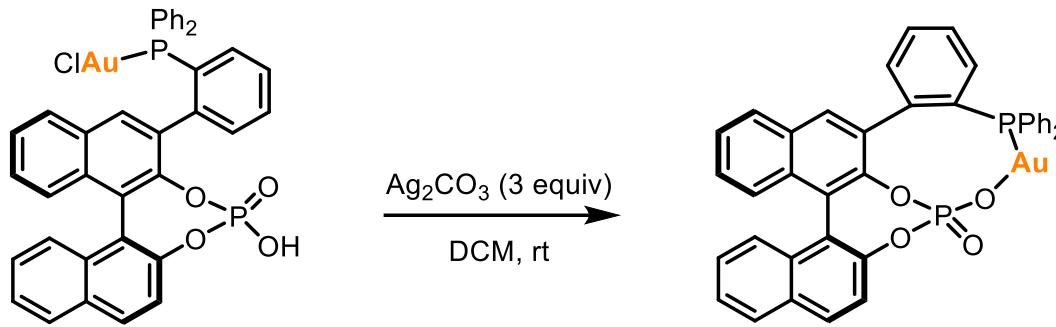


- Straightforward synthesis
- Scalable (3.5 g prepared)!



Silver-mediated activation of the chiral Au(I) catalyst

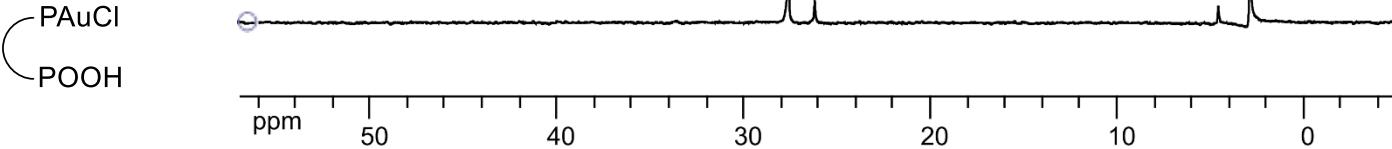
Zhenhao Zhang



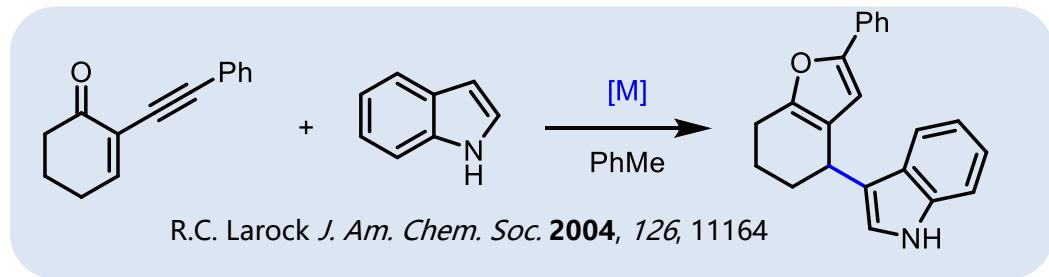
With AgNTf_2



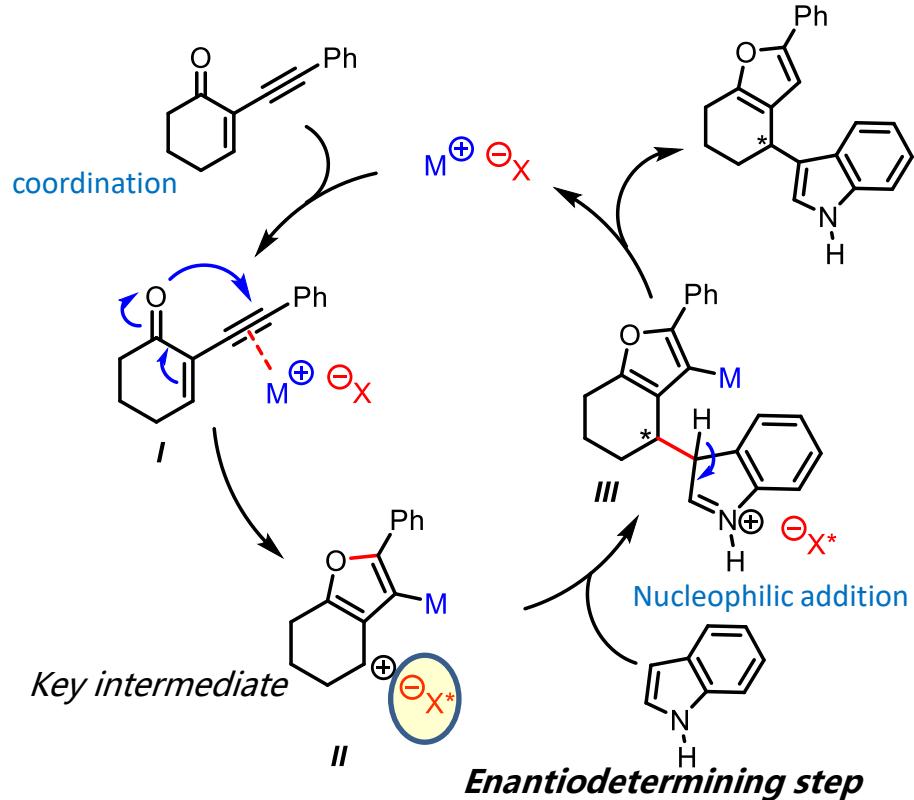
With Ag_2CO_3



Tandem cycloisomerization/nucleophilic addition sequence

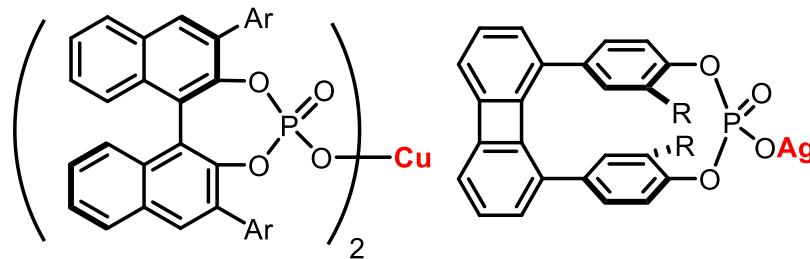


Mechanism



ACDC

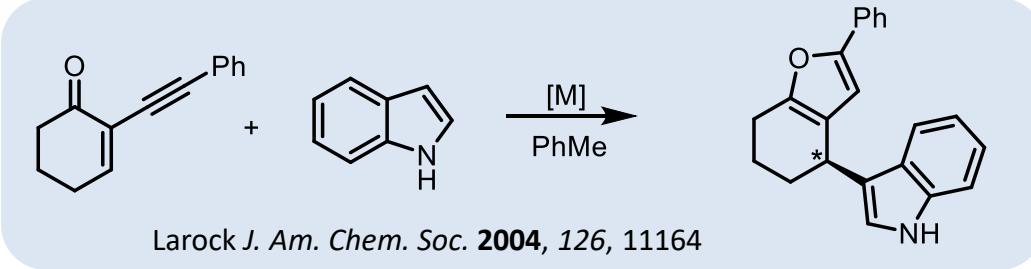
Previous enantioselective variants



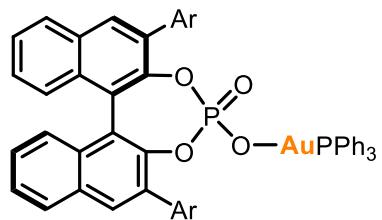
Toste *et al.* *J. Am. Chem. Soc.*, 2011, 133, 8486

Betzer *et al.* *Adv. Synth. Catal.*, 2018, 360, 3356

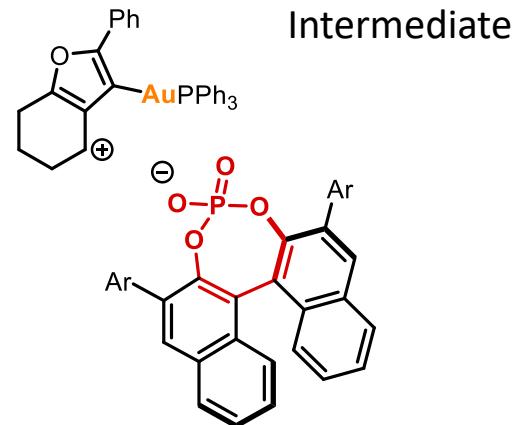
Optimization



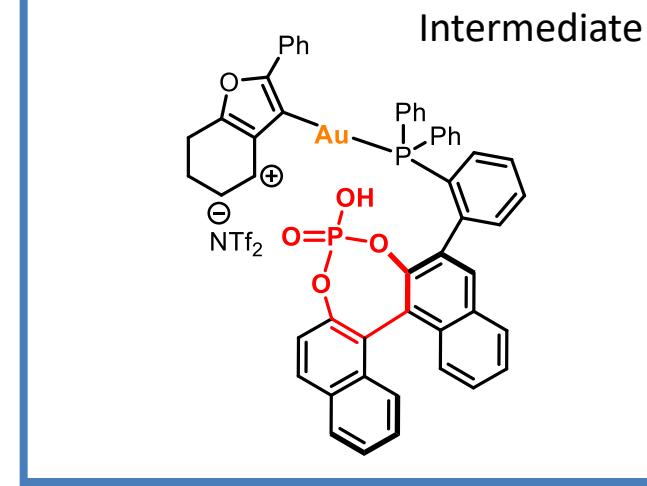
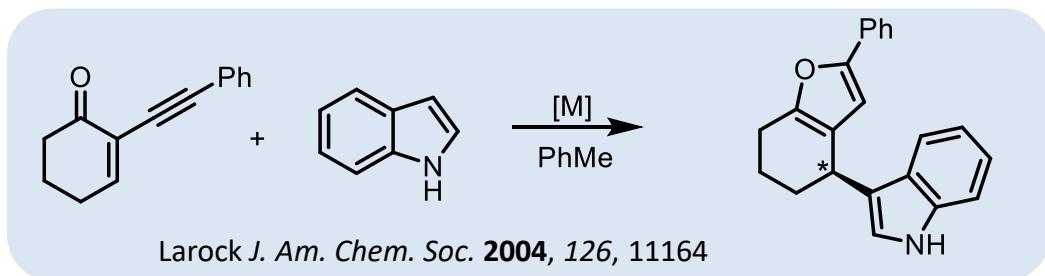
ACDC



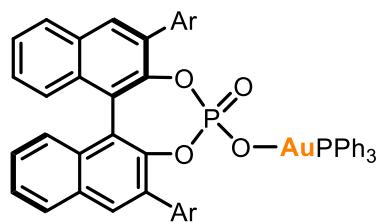
60%, 24% ee



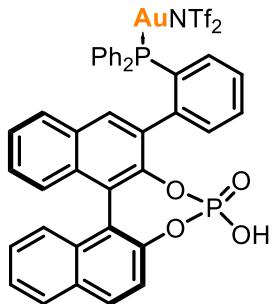
Optimization



ACDC

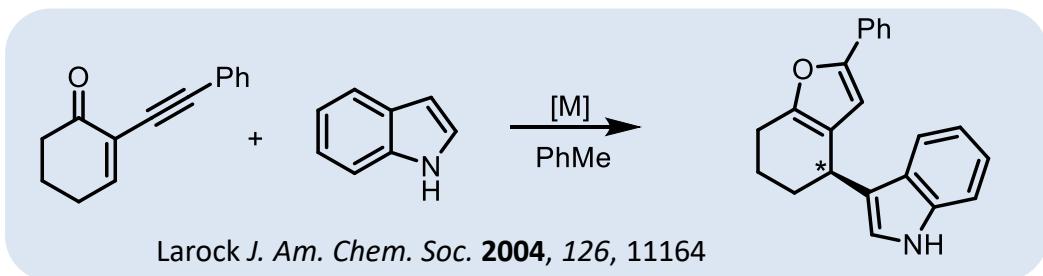


60%, 24% ee

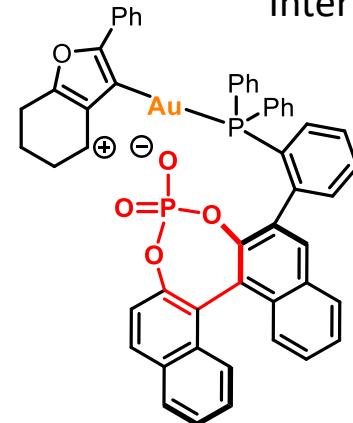


85%, 15% ee

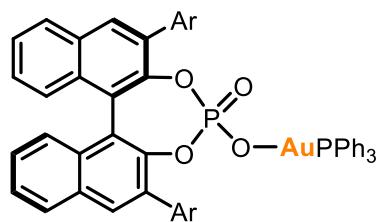
Optimization



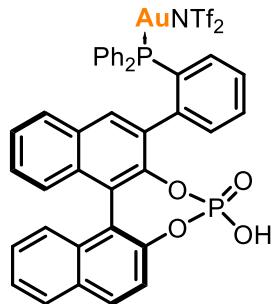
Intermediate



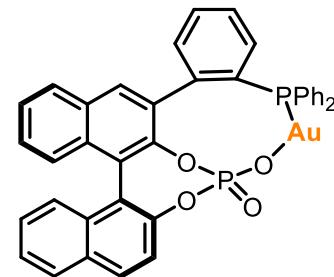
ACDC



60%, 24% ee



85%, 15% ee



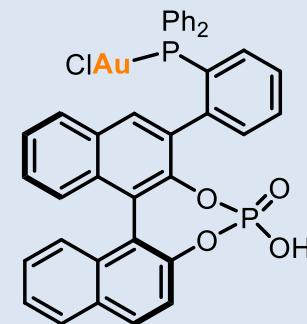
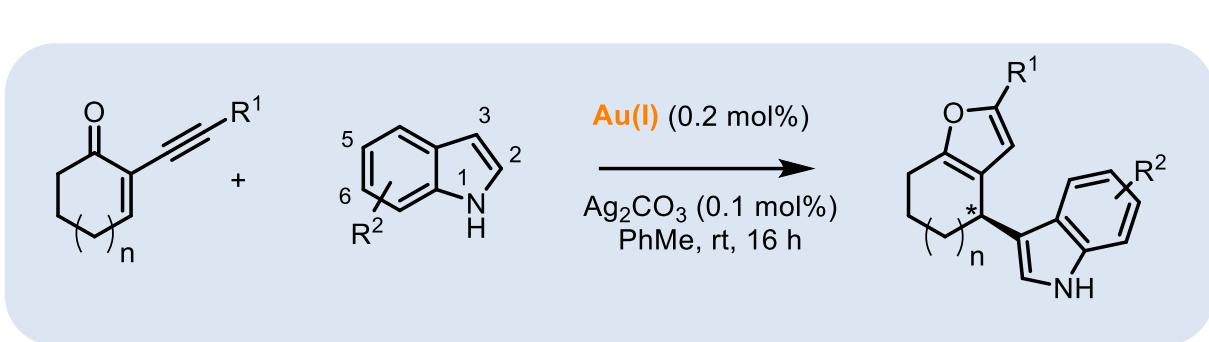
TcDC

75%, 96% ee (0.2 mol%)

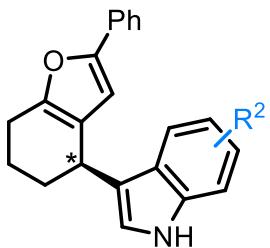
73%, 93% ee (0.1 mol%)

Scope of the reaction

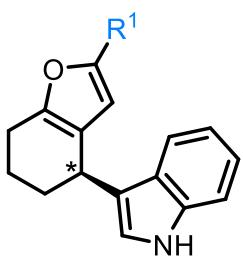
Zhenhao Zhang



C3-alkylation

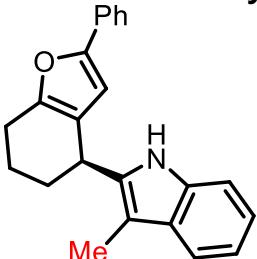


- $R^2 = H$, 75%, 96% ee
- $R^2 = 5\text{-OMe}$, 81%, 97% ee
- $R^2 = 5\text{-Br}$, 77%, 97% ee
- $R^2 = 6\text{-Me}$, 44%, 96% ee
- $R^2 = 2\text{-Me}$, 49%, 82% ee

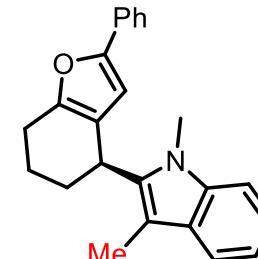


- $R^1 = 4\text{-MeOC}_6\text{H}_4$, 87%, 91% ee
- $R^1 = 4\text{-MeC}_6\text{H}_4$, 95%, 95% ee
- $R^1 = 3\text{-MeOC}_6\text{H}_4$, 77%, 84% ee
- $R^1 = C_5\text{H}_{11}$, 46%, 87% ee

C2-alkylation

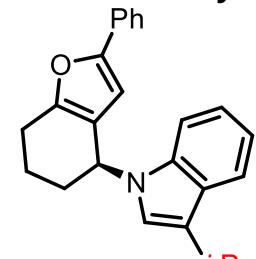


83%, 93% ee



91%, 95% ee

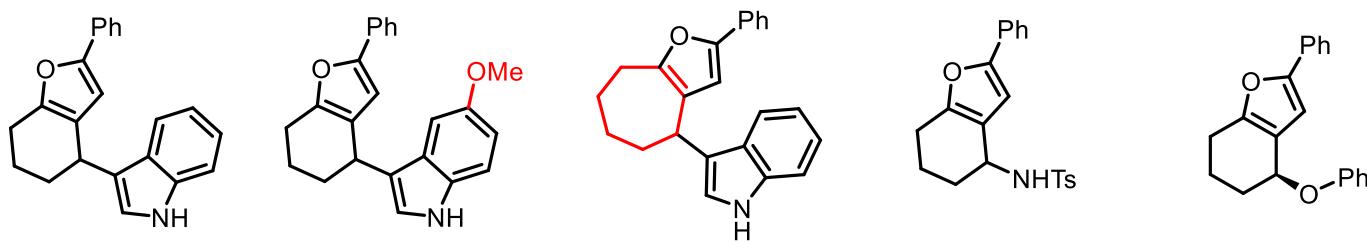
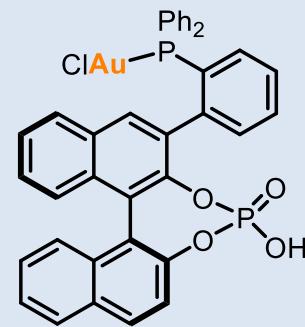
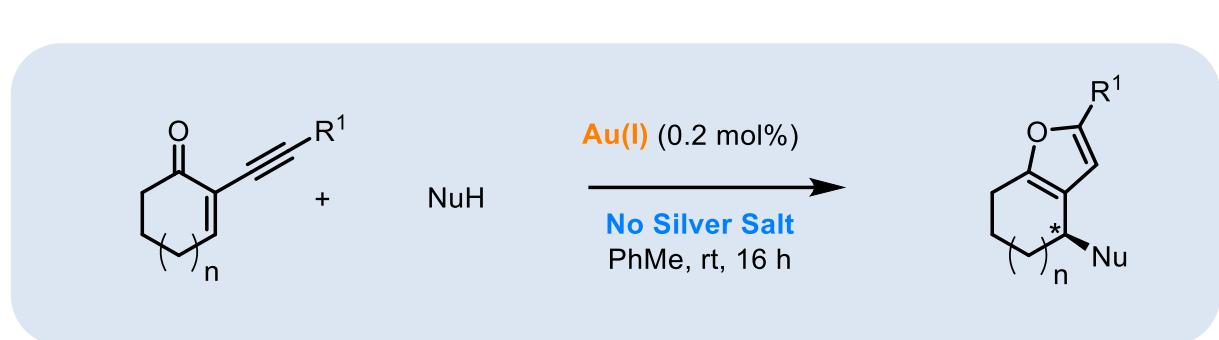
N-alkylation



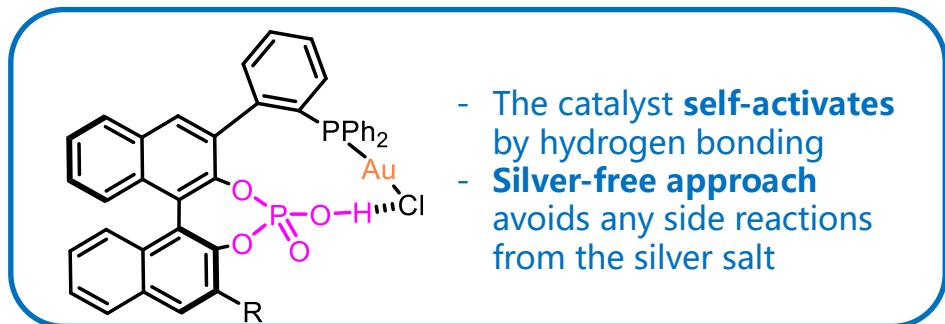
59%, 92% ee

Scope of the reaction: silver-free protocol

Zhenhao Zhang



No Ag_2CO_3	80%, 92% ee	73%, 94% ee	55%, 95% ee	41%, 88% ee	Silver-free: 93%, 90% ee
With Ag_2CO_3	(75%, 96% ee)	(81%, 97% ee)	(41%, 94% ee)	(80%, 87% ee)	

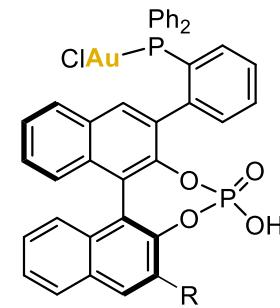
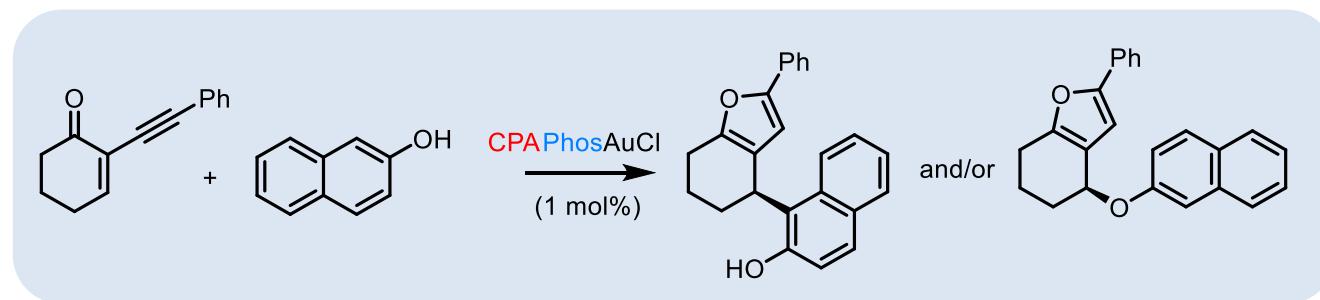


Ag effects in Au(I) catalysis: Shi et al. *J. Am. Chem. Soc.* **2012**, 134, 9012.

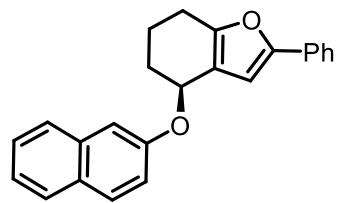
Ag-free Au(I) catalysis: Echavarren et al. *Bull. Chem. Soc. Jpn.* **2021**, 94, 1099 and *Chem. Eur. J.* **2021**, 27, 11989.

Silver-free addition of naphthols

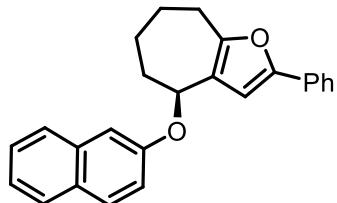
Yunliang Yu
Nazarii Sabat



O-addition products

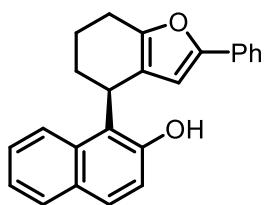


■ (S), 51%, 85% ee
■ (R), >90%, 85% ee (-)^a

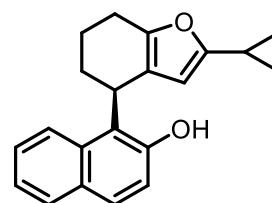


■ (-), 72%, 85% ee
■ (-), 53%, 69% ee

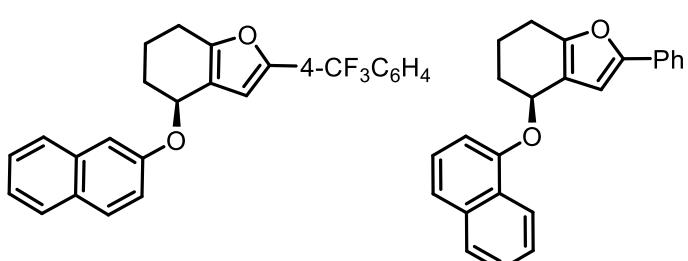
C-addition products



■ 60%, 93% ee

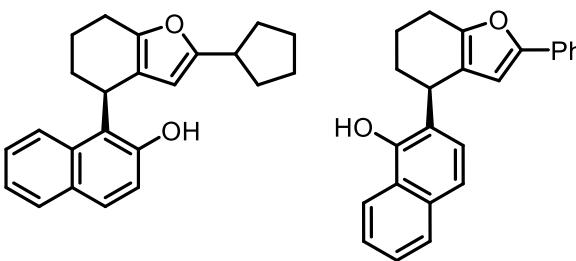


■ 85%, 92% ee



■ (-), 49%, 64% ee

■ (-), 36%, 88% ee
■ (+), 55%, 50% ee

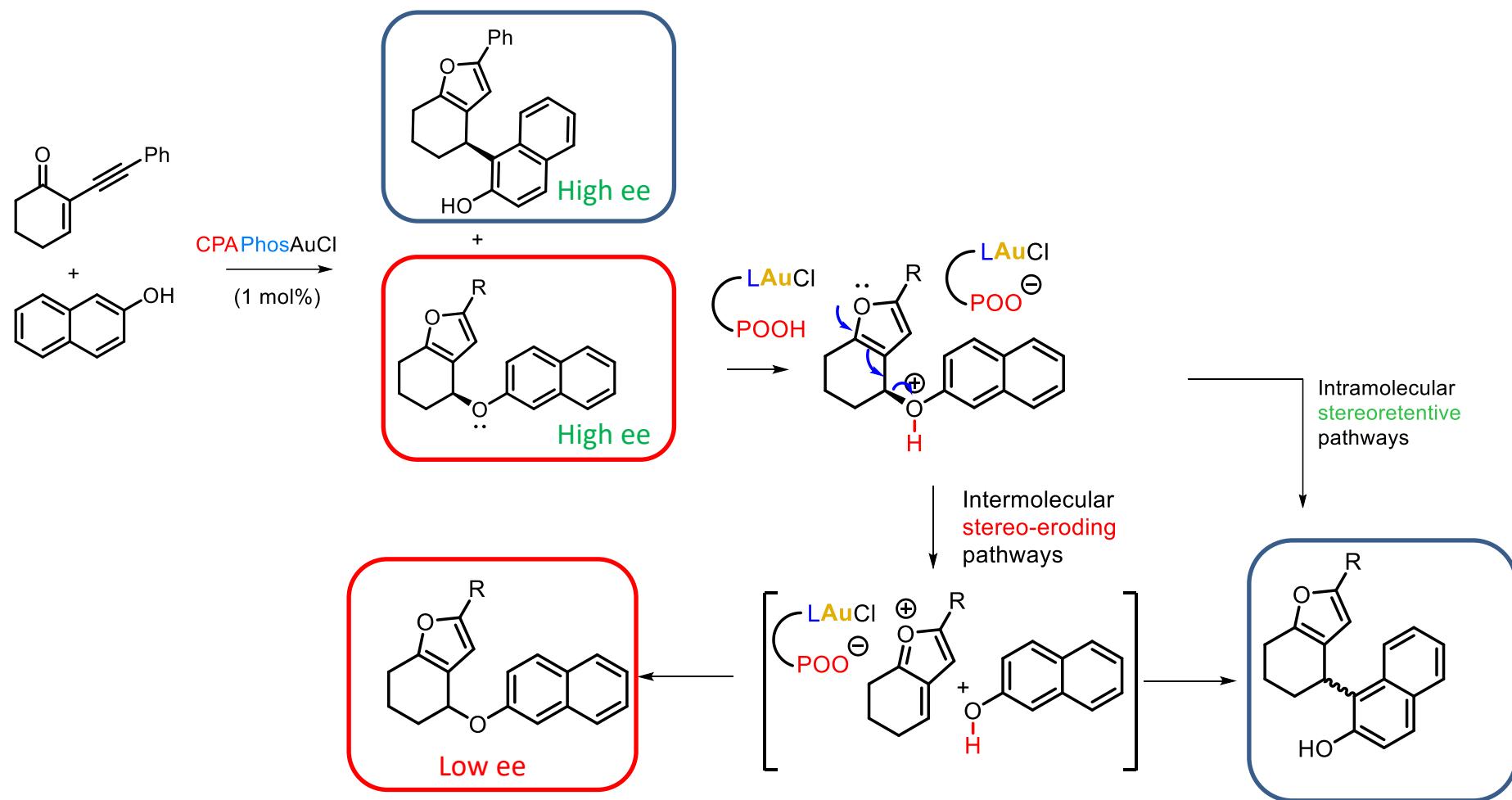


■ 68%, 99% ee

■ 61%, 98% ee

■ $(\text{CPA Phos}^{\text{A}})\text{AuCl}$
■ $(\text{CPA Phos}^{\text{B}})\text{AuCl}$
■ $(\text{CPA Phos}^{\text{B}})\text{AuCl}$

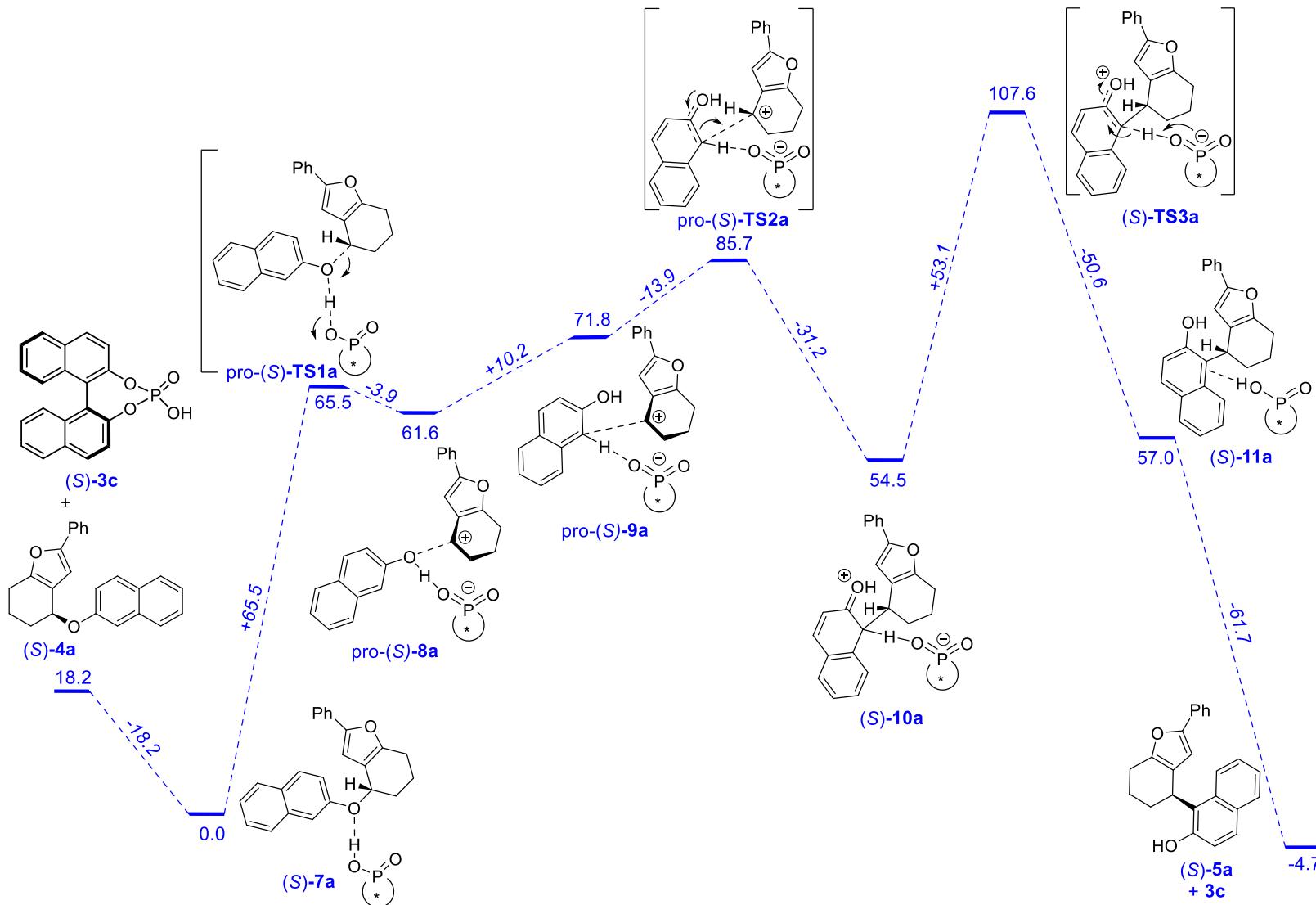
Mechanism of the stereo-eroding pathway



→ The final ees of both products depends in the end on the reaction time and the R group

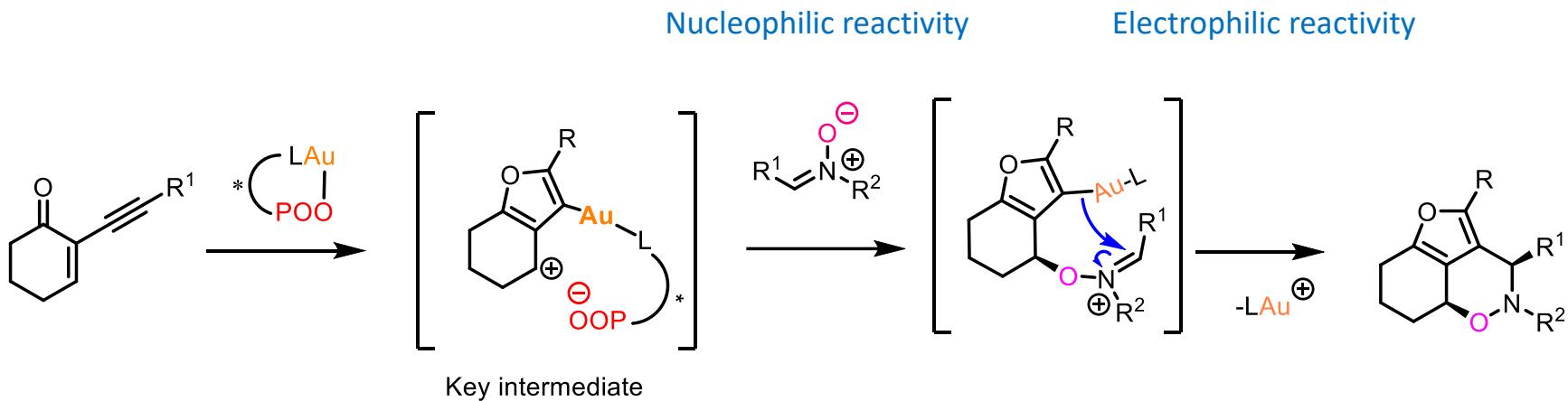
Structural instability: conversion of O-add to C-add products

Coll. with G. Frison

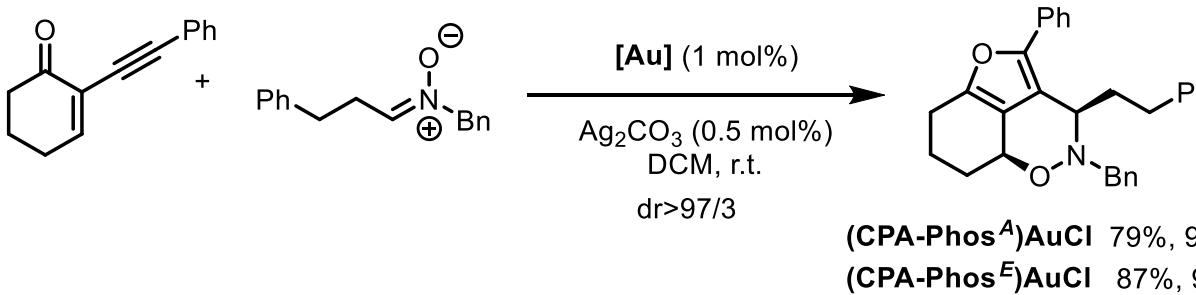


Tandem cycloisomerization/formal cycloaddition sequence

Zhenhao Zhang

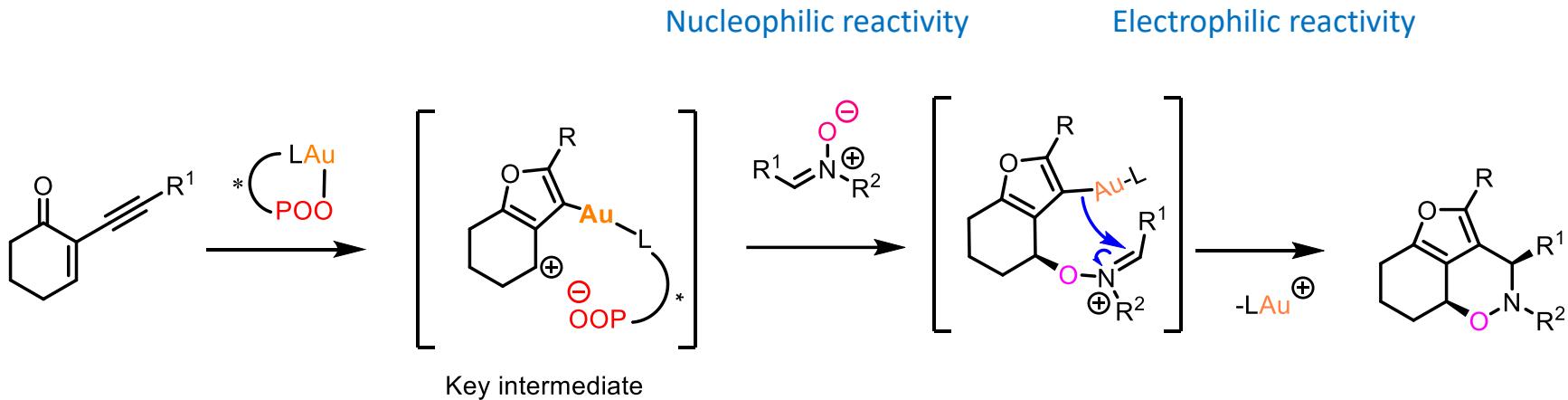


From nitrone cycloadditions...



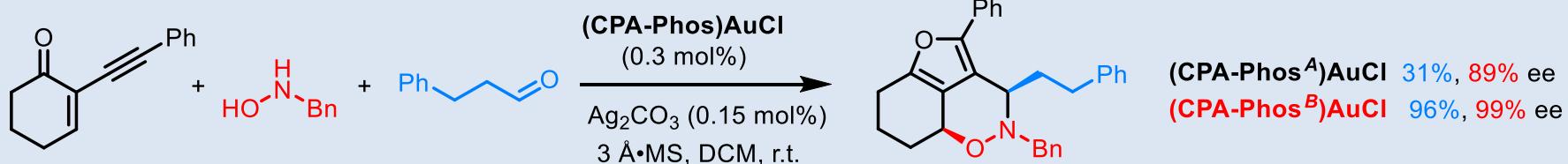
Tandem cycloisomerization/formal cycloaddition sequence

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From nitrone cycloadditions...

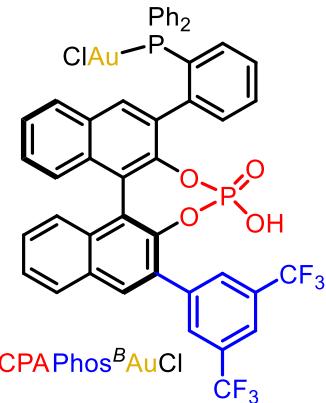
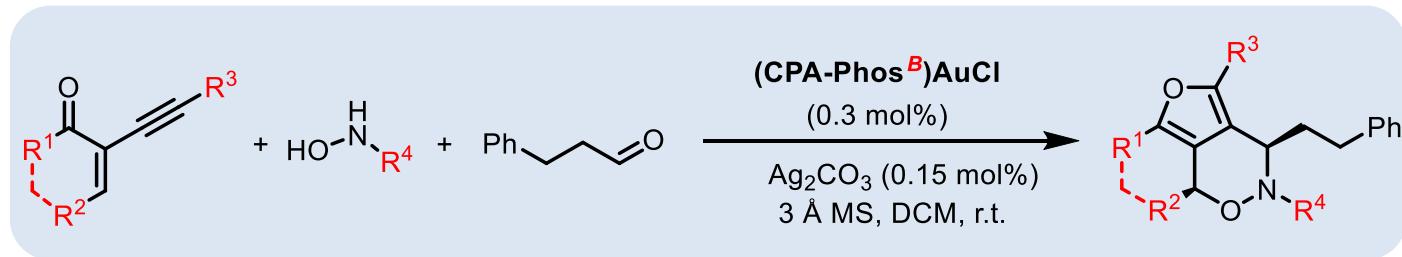
... to a multicomponent version of the reaction!



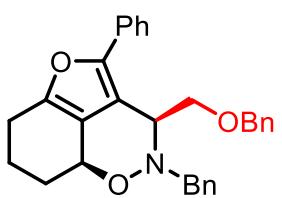
→ $(CPA\text{-Phos}^B)\text{AuCl}$ shows a higher performance
→ MCR version avoids the synthesis of the nitrone

Scope of the multicomponent reaction

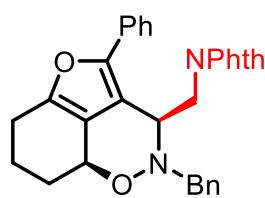
Zhenhao Zhang



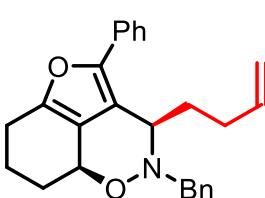
Aliphatic aldehydes



85%, 99% ee

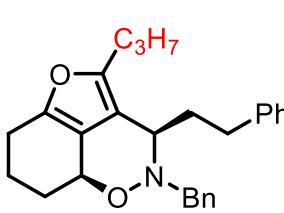


97%, 99% ee

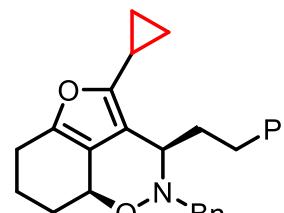


61%, 97% ee

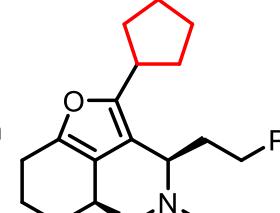
Alkyl R³ substituent



44%, 89% ee

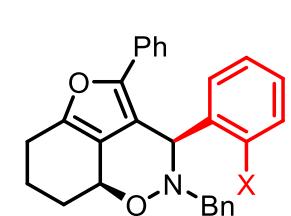


58%, 92% ee

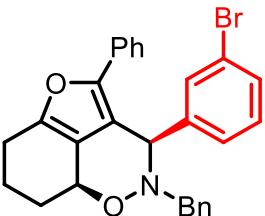


63%, 90% ee

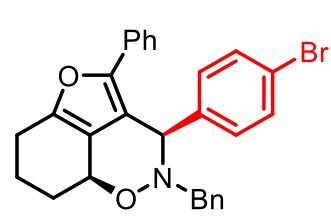
Aromatic aldehydes



X=Br, 48%, 99% ee
X=F, 70%, 97% ee

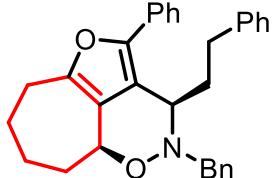


87%, 92% ee

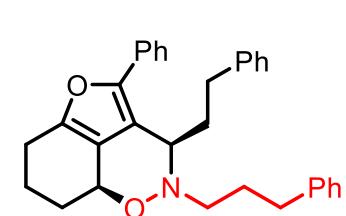


75%, 99% ee

7-membered ring



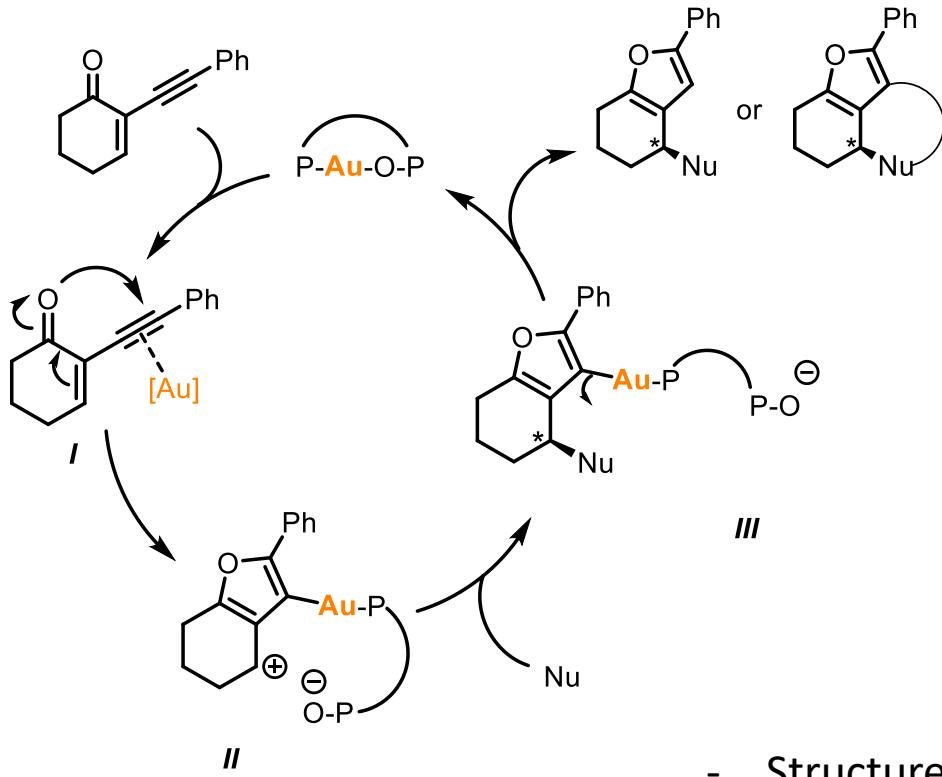
39%, 97% ee



86%, 99% ee

Other hydroxylamines

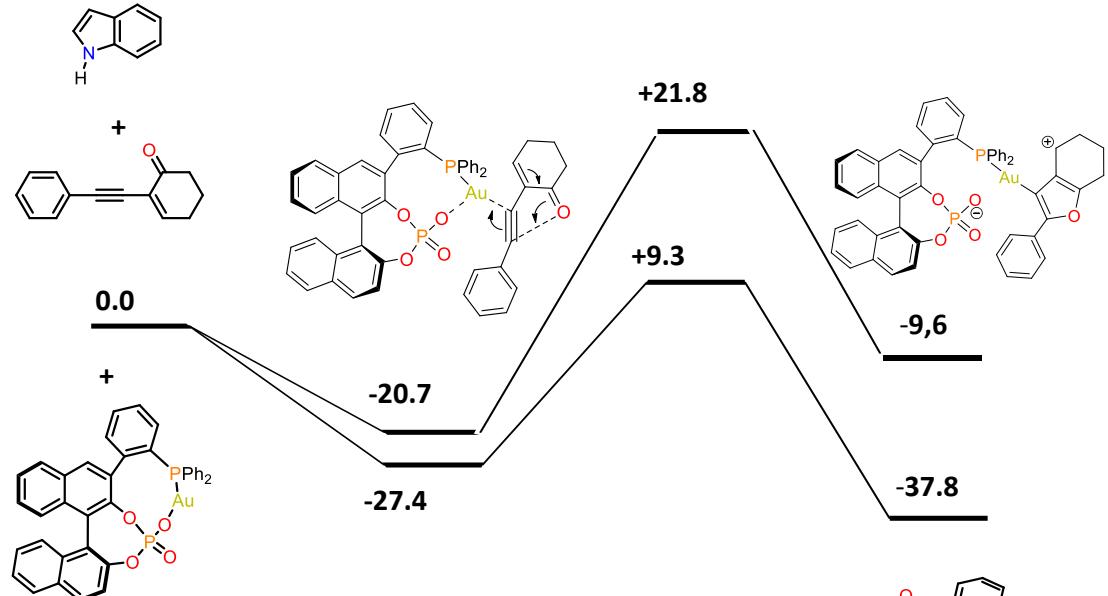
Mechanisms and DFT Calculations



- Structure of the actual catalysts
- Stereoselectivity

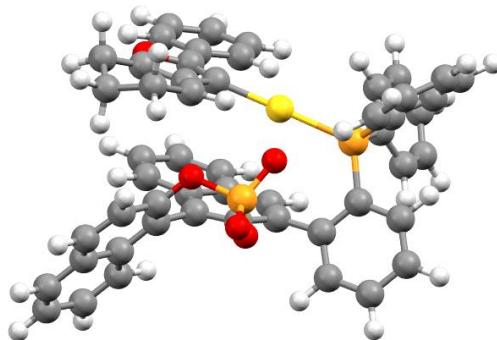
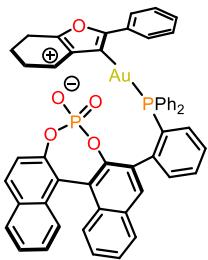
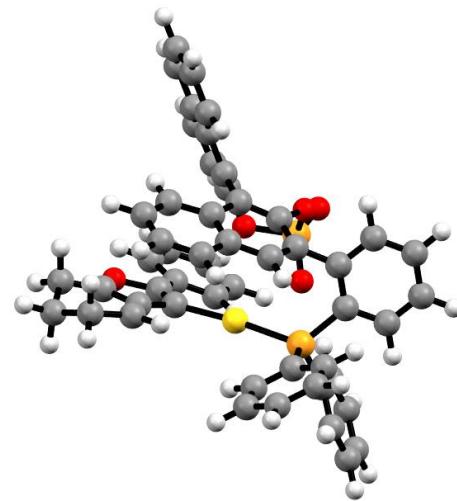
DFT Calculations: enantioselectivity

Coll. with G. Frison



coordination

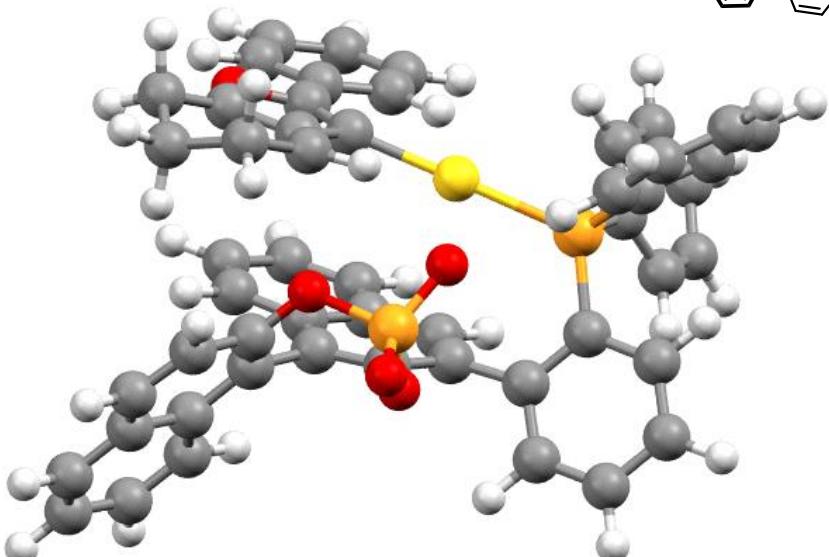
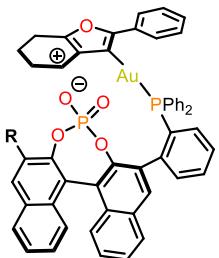
cycloisomerization



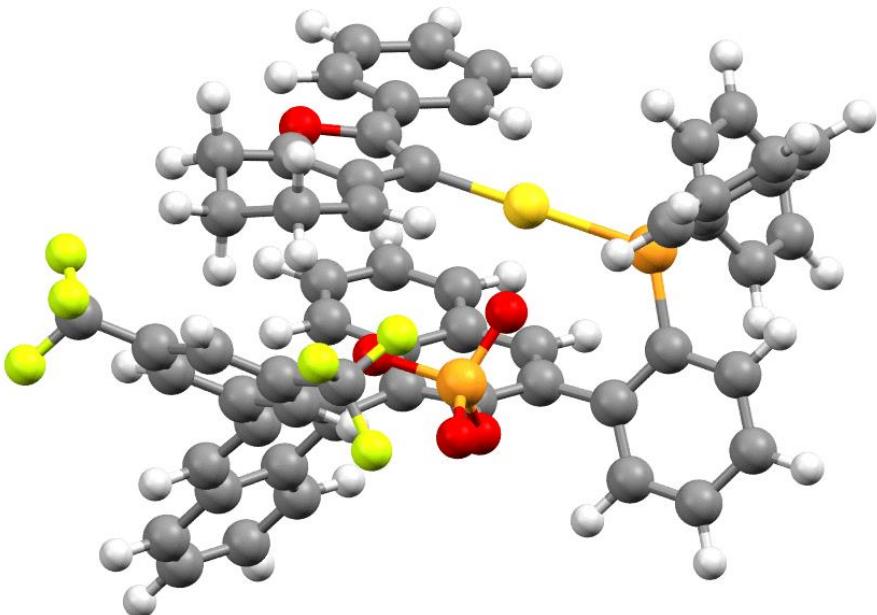
DFT calculations: Carbocation intermediates

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With (CPA-Phos^A)AuCl



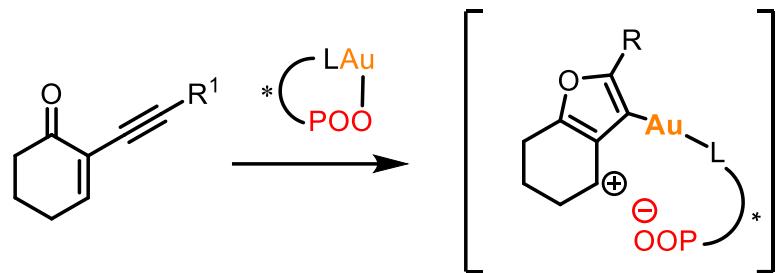
With (CPA-Phos^B)AuCl



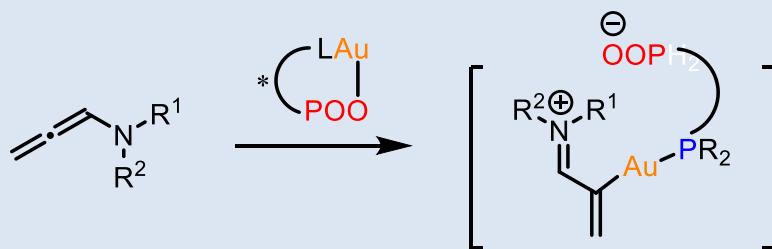
IEPCM(DCM)-M06/def2-TZVPP

- Both intermediates display similar geometries
- The fluorinated aryl group has little impact on the enantioselectivity
- The presence of a hydrogen bond may result in a better charge separation

New modes of activation for TCDC strategy



Activation of 2-alkynyl ketones

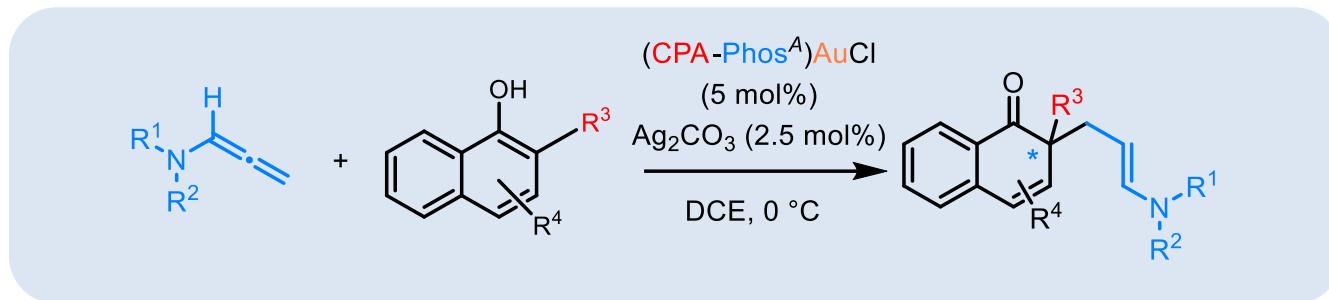


Activation of allenamides

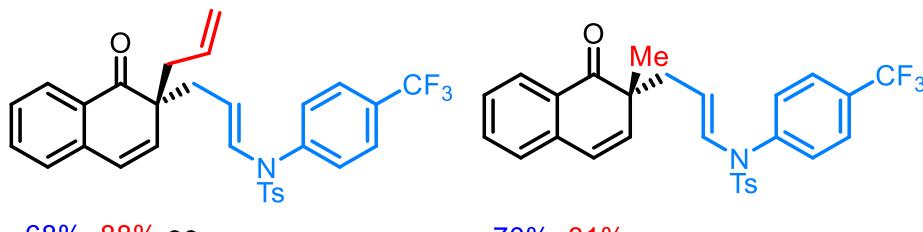
Chem. Commun. **2021**, 57, 10779.

Dearomatization of naphthols with allenamides

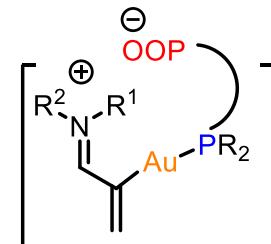
Yunliang Yu



Representative examples



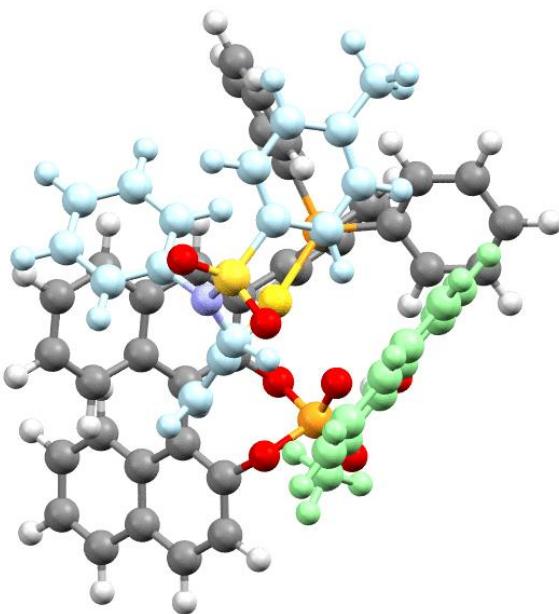
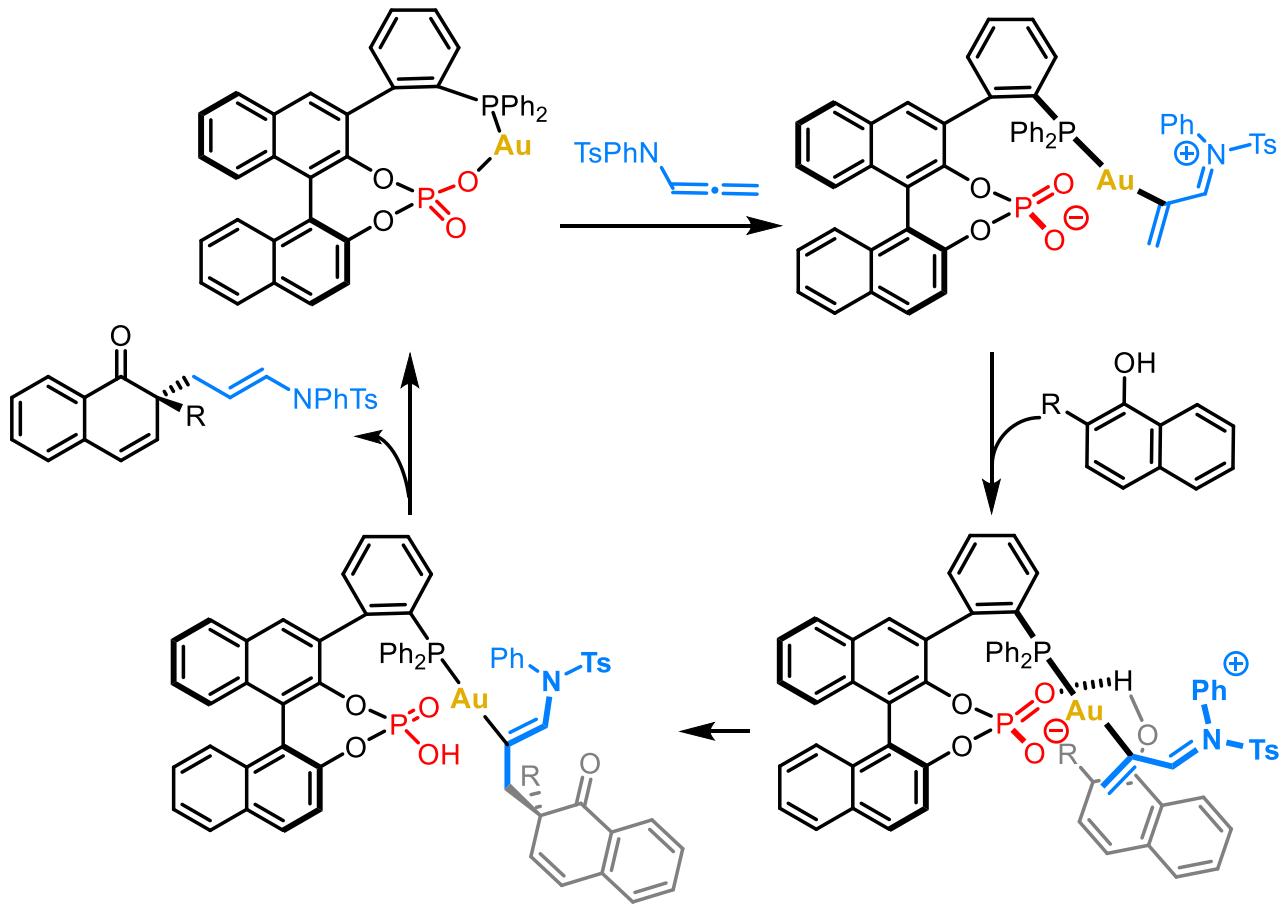
Intermediate



- Dearomatisation of naphthols with allenamides
- Applies to both 1- and 2-naphthols

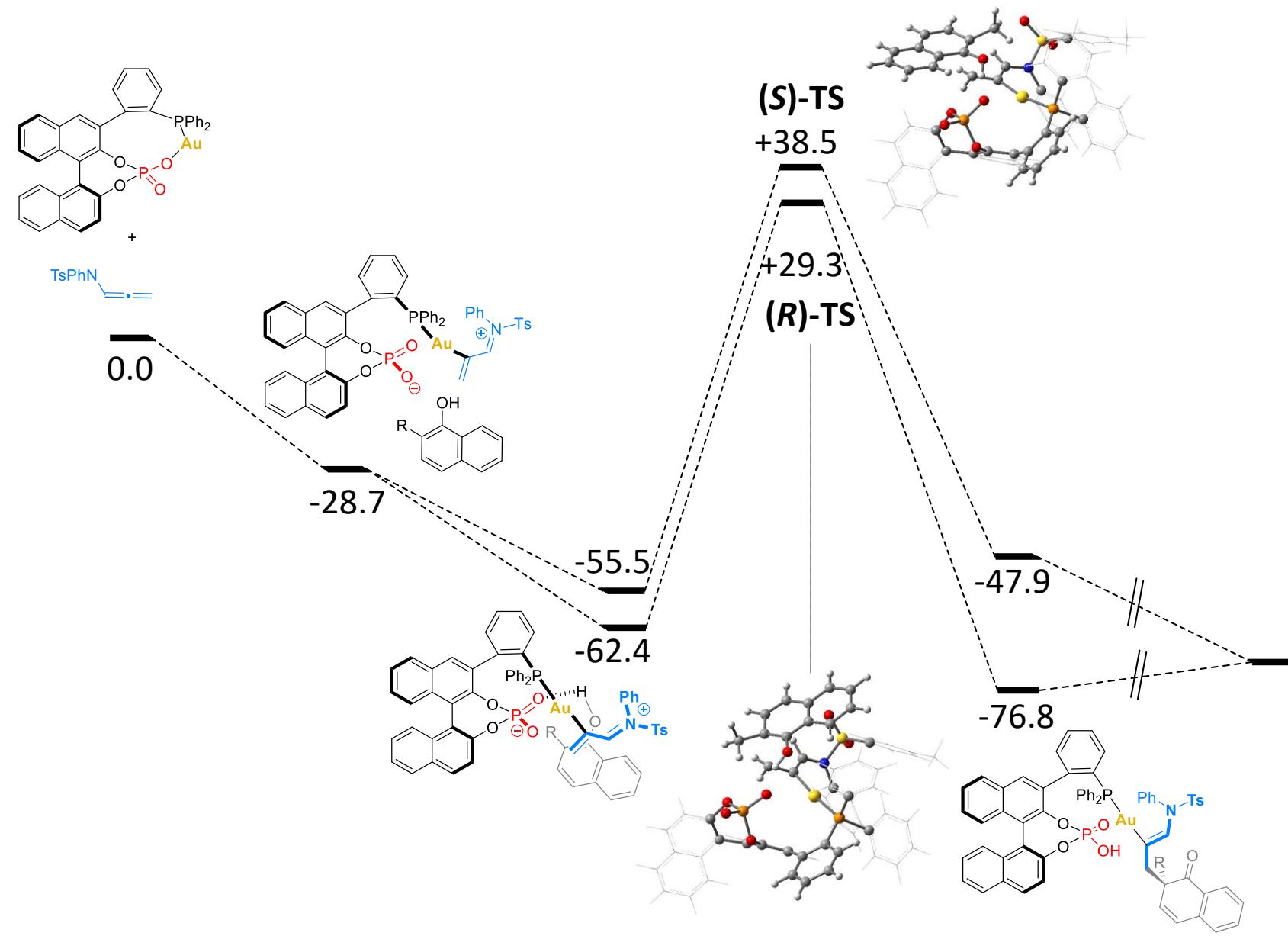
Mechanistic hypothesis

Coll. with G. Frison



DFT calculations

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Acknowledgements



Zhenhao Zhang

Nazarii Sabat

Meriem Daghmoum

Hao Xu

Yunliang Yu

Pengyu Zhou

Mohammed Ramdani



Collaborations
Gilles Frison(LCT, Paris)
Olivier Berteau (INRA)
Olivier Baslé (LCC)