

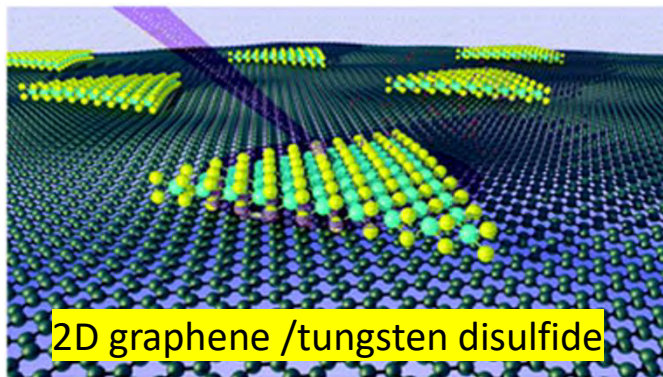
Tunable Photoemission and Photocatalytic Activity of Au-ZnO Nanostructures

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Photocatalysis is powerful, sustainable and cheap technology for energy production and environment remediation

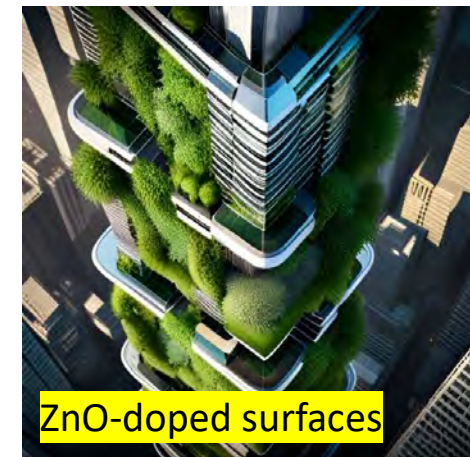
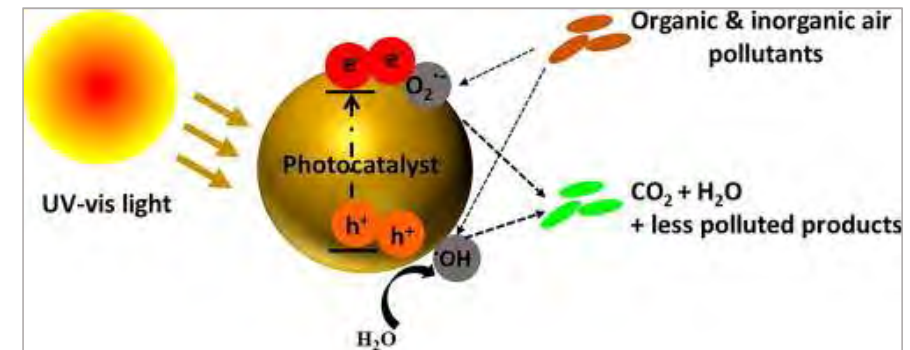
Energy production, storage and conversion



Si-doped Li batteries



Pollutant degradation



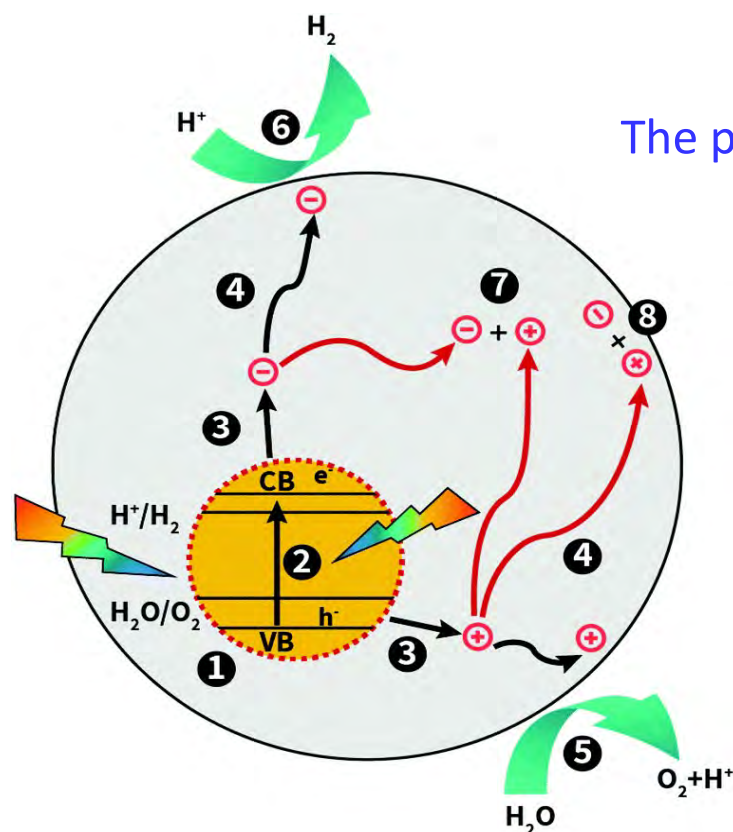
- Pham, H. D. et al. (2021). *Materials Letters*, 288, 129355.

- Hoang, A. T. et al. (2023). *ACS Sustainable Chemistry & Engineering*, 11(4), 1221-1252.

Limitations for photocatalysts!

A process using a material called photocatalyst that absorbs light energy to promote chemical reactions

Photocatalyst = Semiconductor



The photocatalytic fundamental process :

- 1 Light irradiation
- 2 Electron-hole generation
- 3 Electron-hole separation
- 4 Electron-hole transportation to the surface
- 5 Oxidation reaction and O₂ generation
- 6 Reduction reaction and H₂ generation
- 7 Bulk recombination
- 8 Surface recombination

Water splitting mechanism

Kumar, A., et al. (2022). Small, 18(1), 2101638.

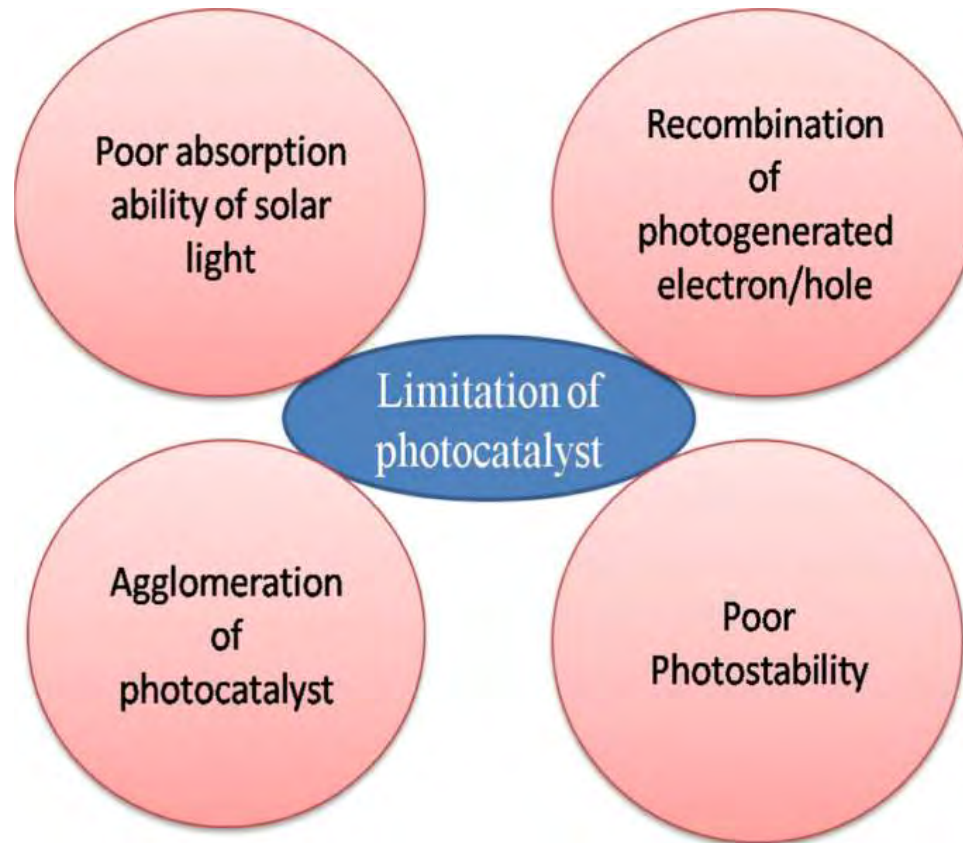
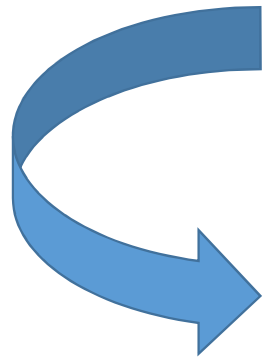
Saravanan, R. et al. (2017). Nanocomposites for visible light-induced photocatalysis, 19-40.

What's challenging in photocatalysis

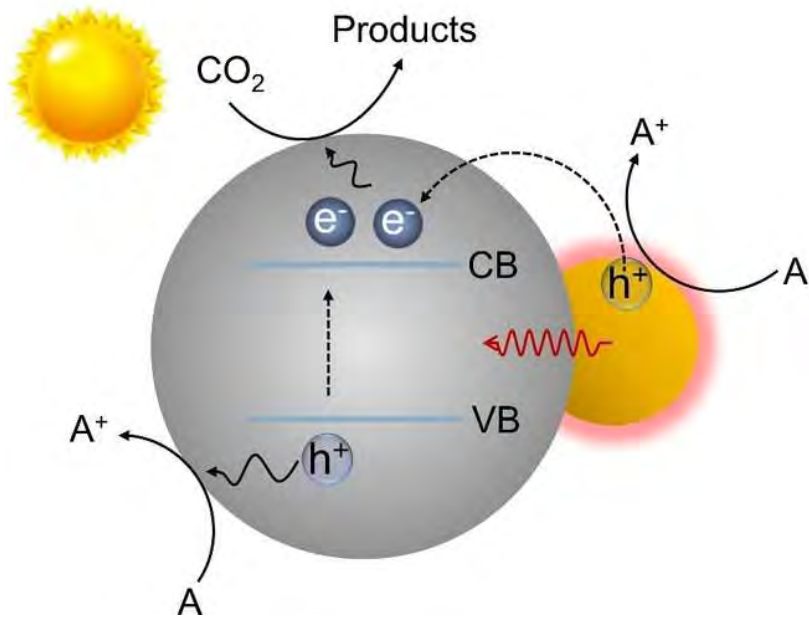
Requirements for a photocatalyst:

- **Chemical reactivity:** robust material with high surface energy and coordination sites
- **Selectivity:** For example selective oxidation of NO to NO₂ and not to another product
- **Stability:** resistance to temperature, surrounding impurities, etc.
- **Activity:** if all previous features are insured, the catalyst is able to promote a reaction
- **Long life span: number of reaction cycles**

Non-homogenous and Defect-rich photocatalyst



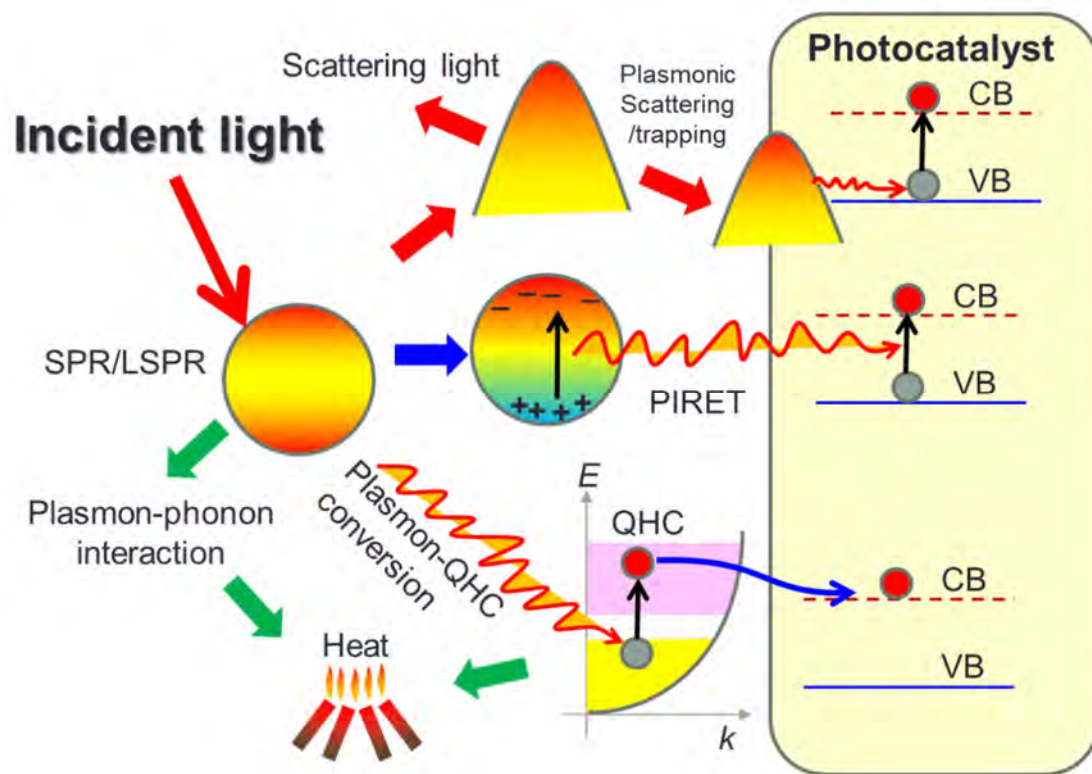
Metal-semiconductor heterojunction



- harness a **broader range of the solar spectrum**, allowing for more efficient use of sunlight.
- offer greater control over reaction **selectivity**
- can operate under **milder conditions**, reducing energy consumption and costs.

Plasmonic photocatalysis: mechanistic issues

The plasmonic energy transformations of hot electrons in photocatalysis



- Light scattering
- Local EF enhancement induced-hot electron injections
- Local heat generation

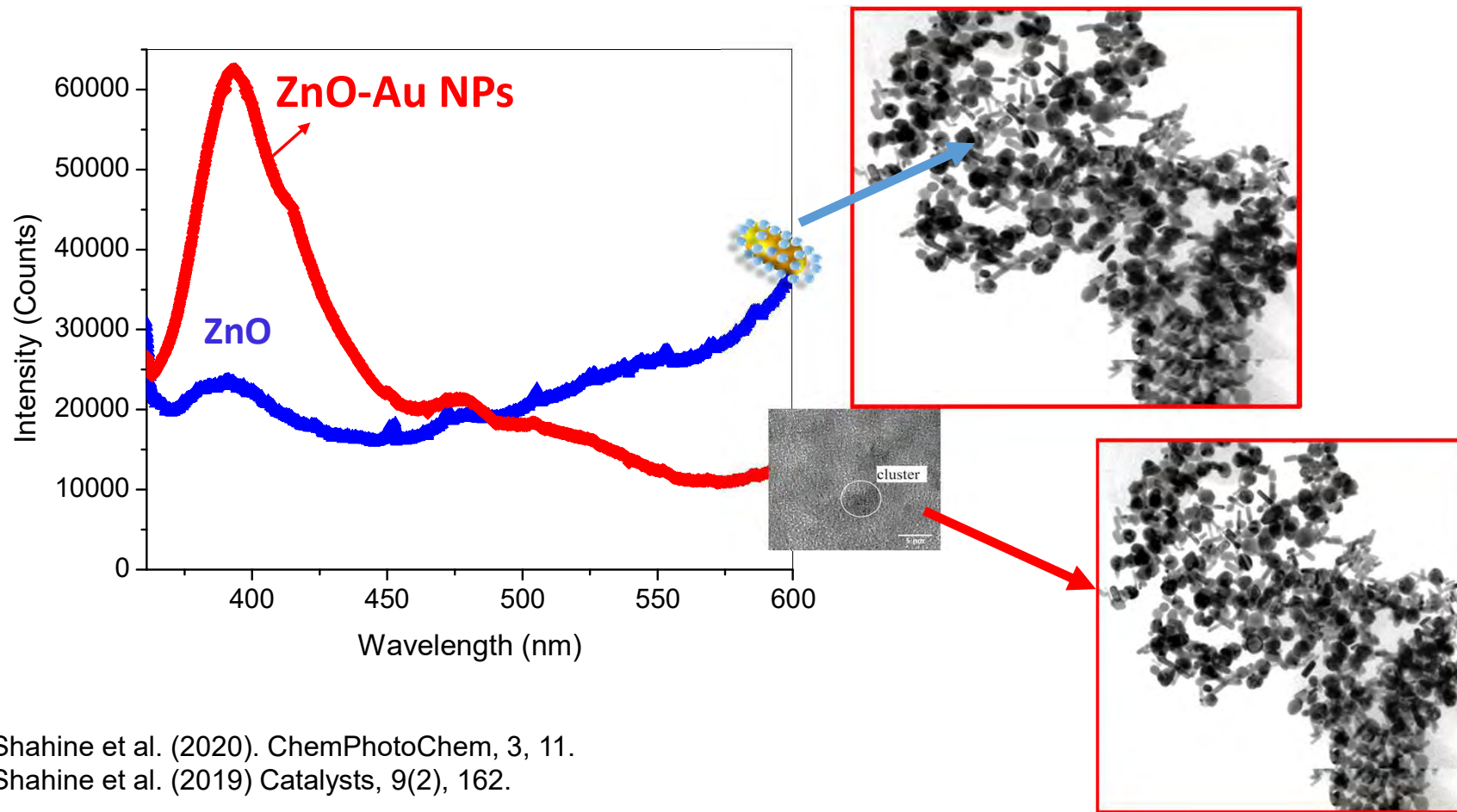
- The dynamics in surface plasmons is still incomplete with information lacking on the detailed mechanism

Lin, Z. et al. (2015). *Nanoscale*, 7(9), 4114-4123.

Shahine, I., Jradi, S., Beydoun, N., Gaumet, J. J., & Akil, S. (2020). *ChemPhotoChem*, 3, 11.

Limitations for nanostructures prepared in liquid medium

Characterization of solutions doesn't allow good understanding of heterojunction mechanisms : need for synthesis optimization

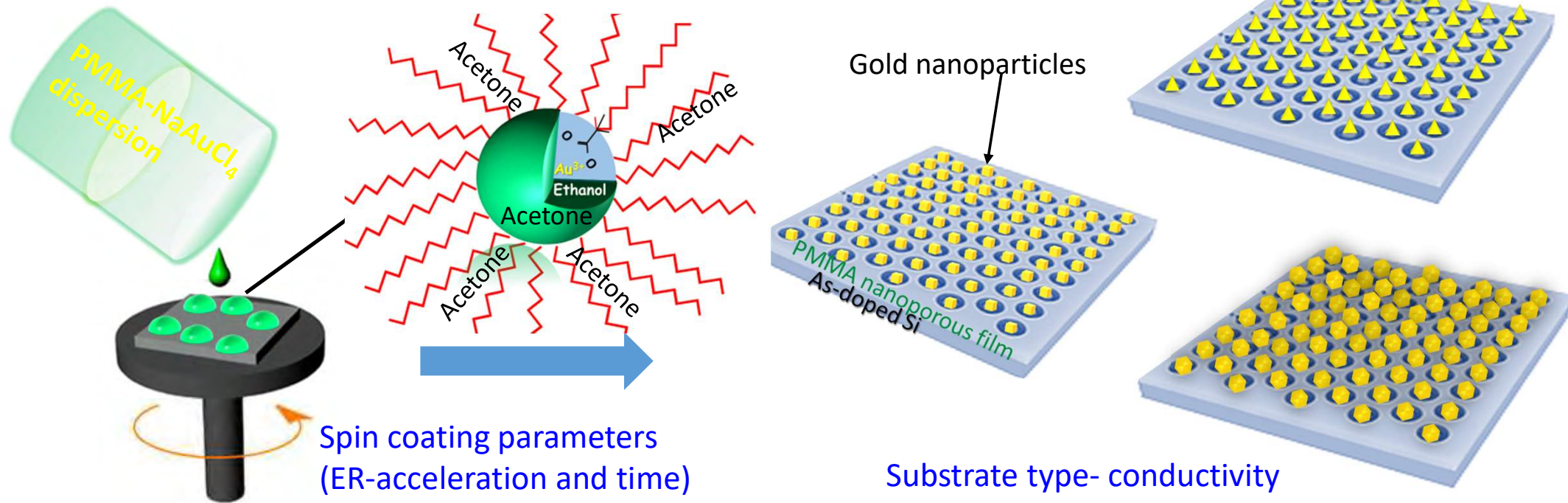


- Shahine et al. (2020). ChemPhotoChem, 3, 11.
- Shahine et al. (2019) Catalysts, 9(2), 162.

Shape-controlled synthesis of gold nanoparticles

Experimental approach

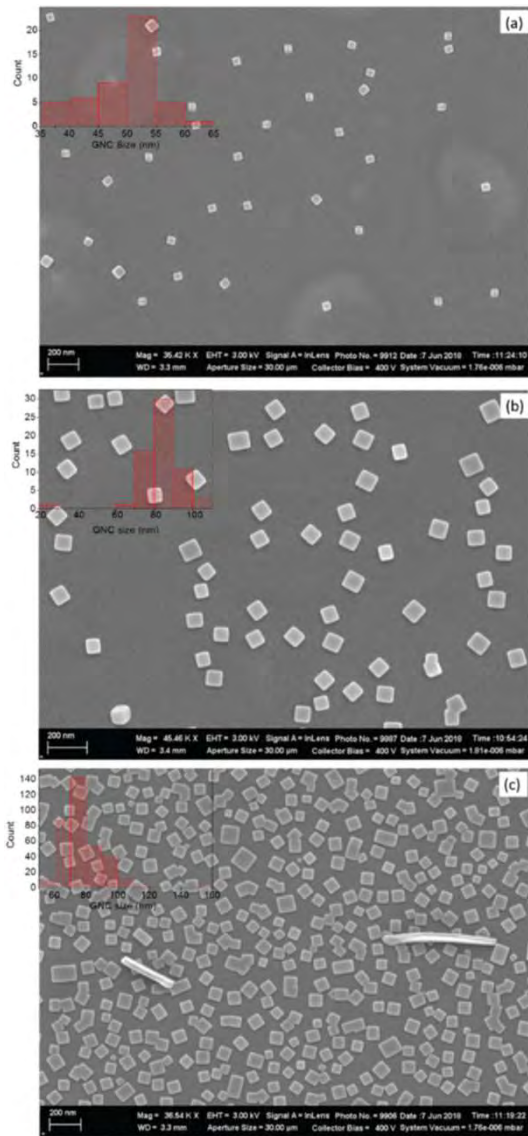
Dispersion composition: PMMA solvent (acetone),
PMMA non-solvent (ethanol)- gold precursor (NaAuCl_4)



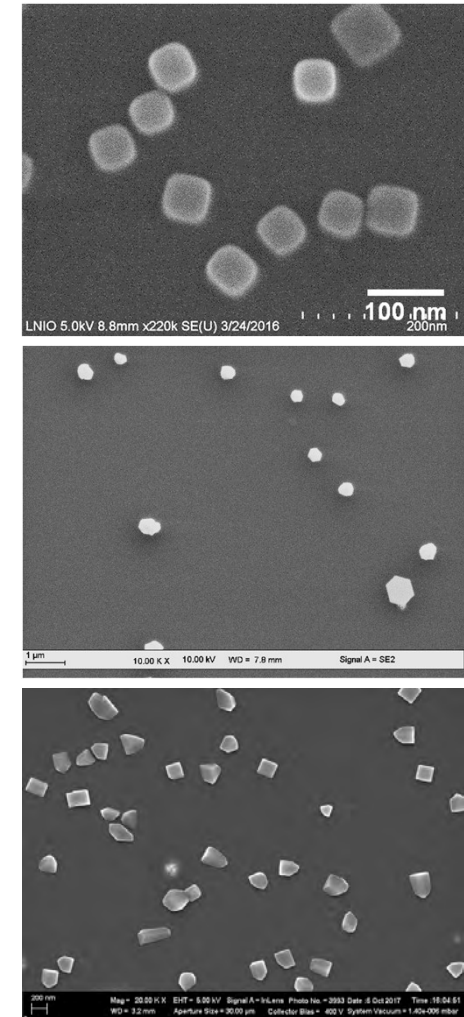
Spin-coating induced the formation of Au^{3+} -loaded PMMA micelles

Vapor induced phase separation allows a spontaneous formation of monodisperse gold nanoparticles

Engineering plasmonic nanomaterials for photocatalysis



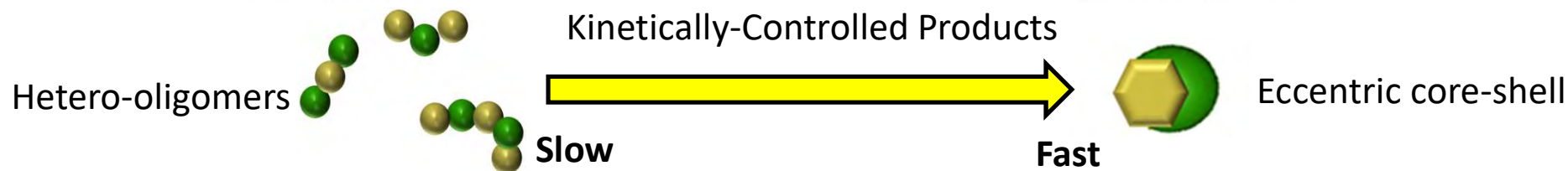
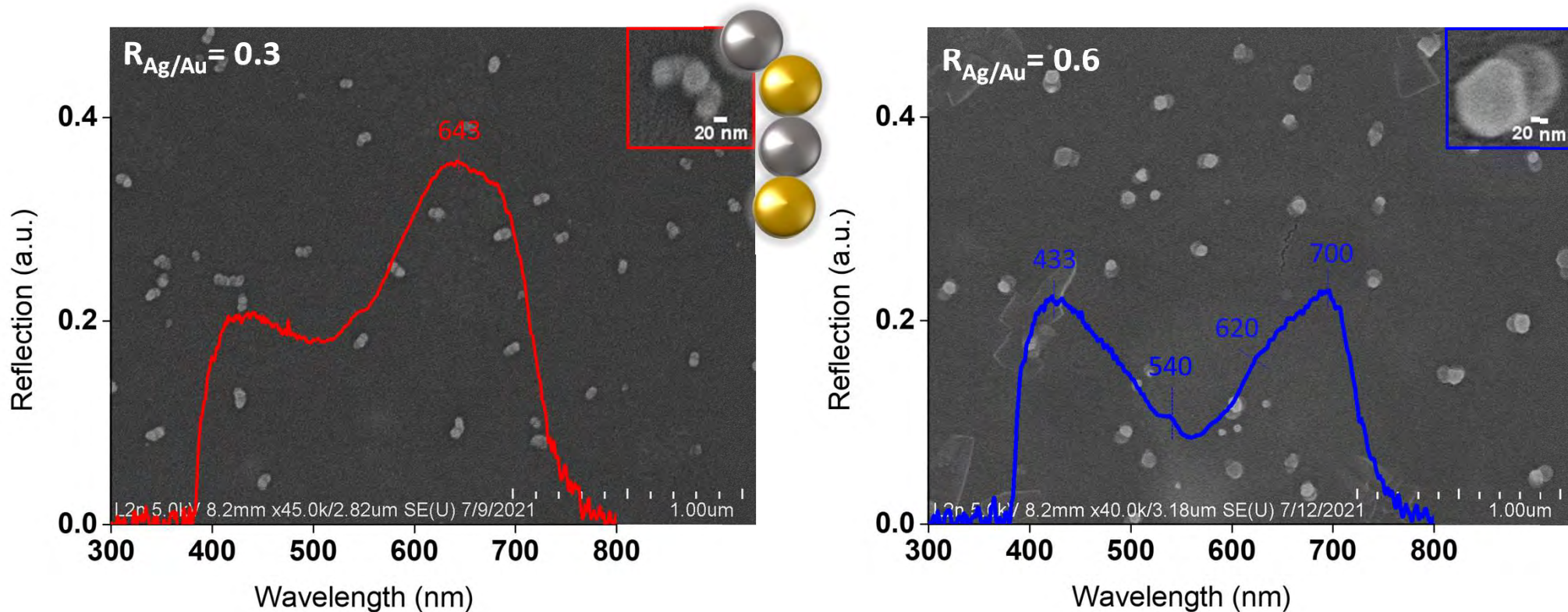
Shape control Gold nanoparticles



-Fahes et al. (2023). *SoftMatter*, 19, 321.

-Akil, S., Omar, R., Kuznetsov, D., Shur, V., En Naciri, A., & Jradi, S. (2021). *Nanomaterials*, 11(7), 1806.

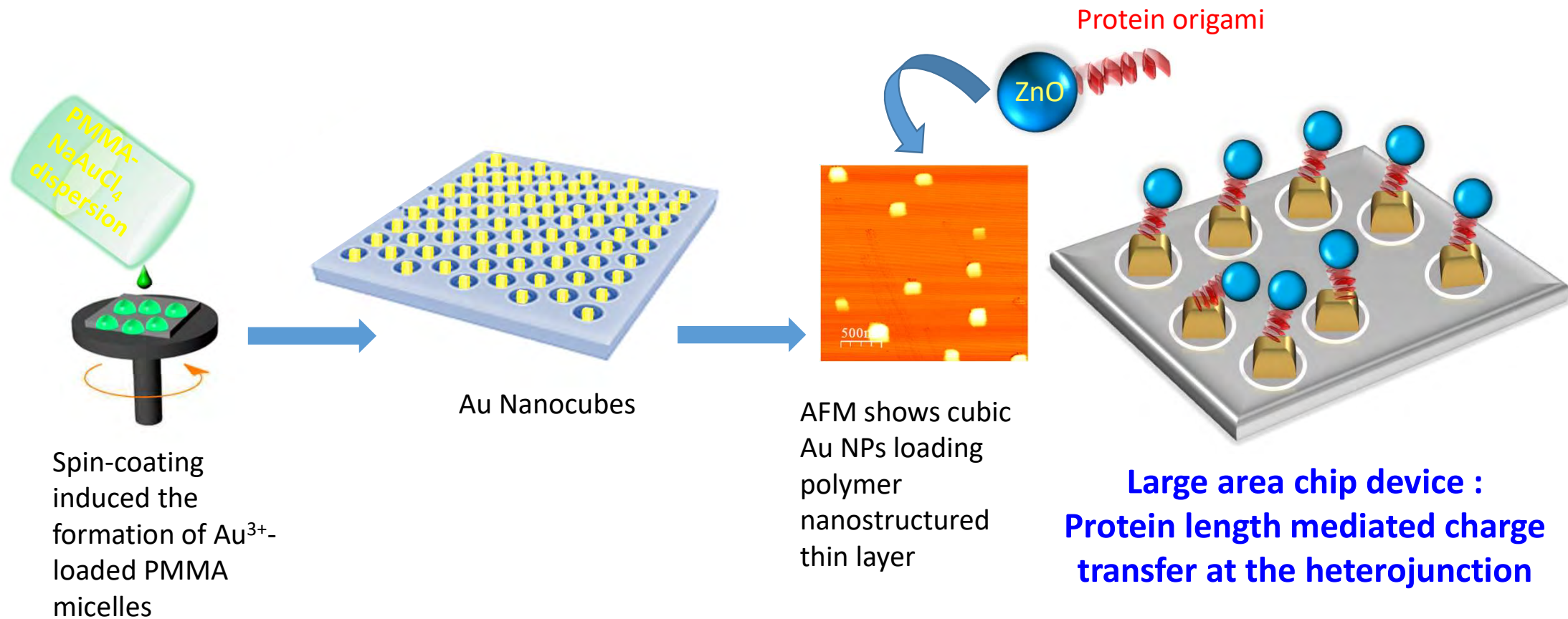
Anisotropic shape-controlled Ag/Au bimetallic nanostructures



-Fahes et al. (2022). Sensors and biosensors Research, 38, 100528.

-Fahes et al. (2021). Nanomaterials, 11(8), 2055.

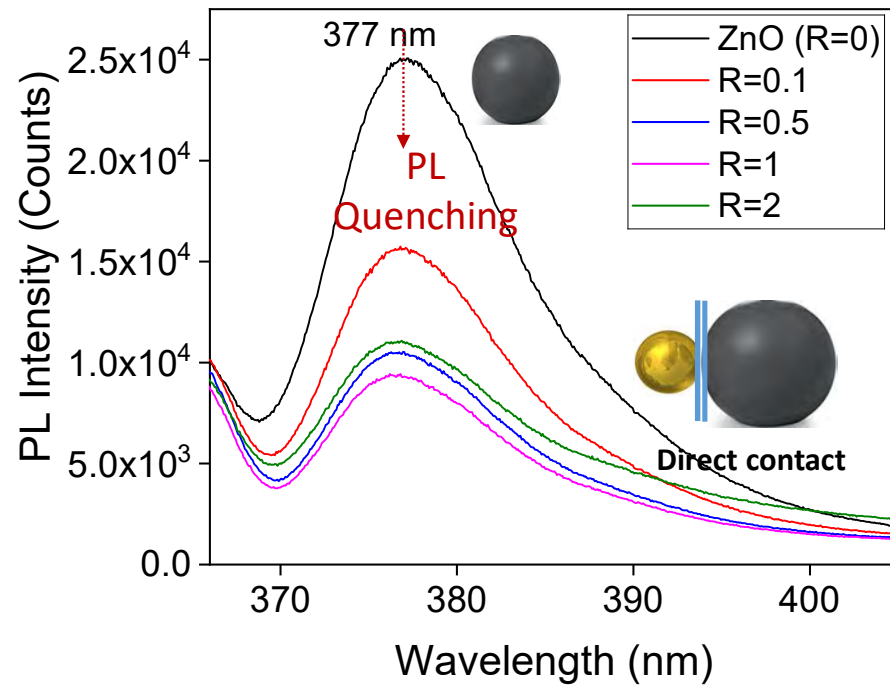
Engineering Au-ZnO nanostructures by a Surface-based synthesis strategy



- Fahes A. et al. (2023). SoftMatter, 19, 321.
- Fahes A. et al. (2022). Sensors and biosensors Research, 38, 100528.
- Akil S. et al. (2021). Nanomaterials, 11(7), 1806.
- Fahes A. et al. (2021). Nanomaterials, 11(8), 2055.

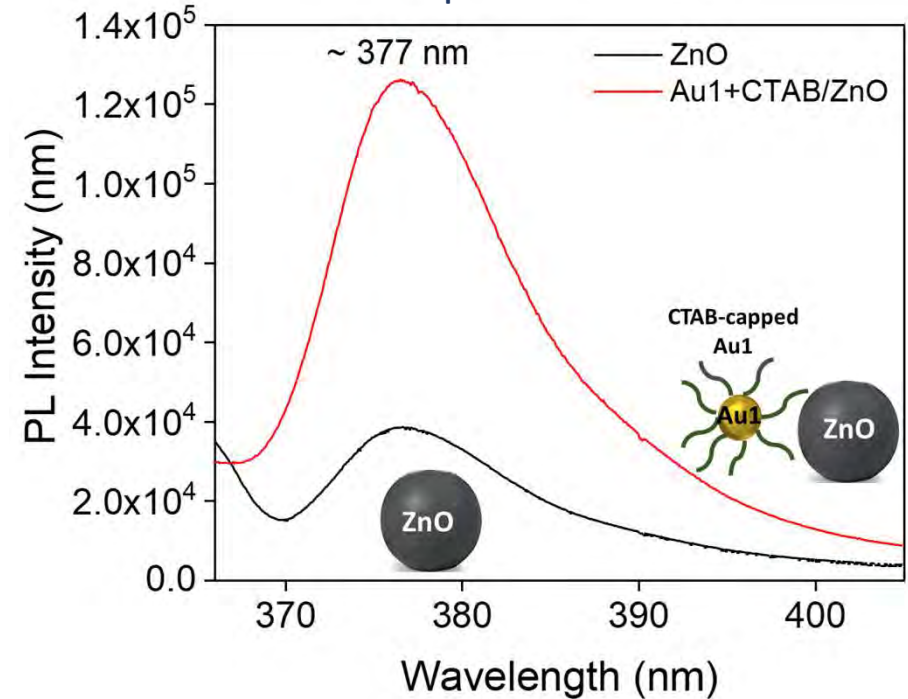
From quenched to enhanced photoluminescence

Pure Au-ZnO NPs



- Au-ZnO junction
- Formation of Schottky barrier

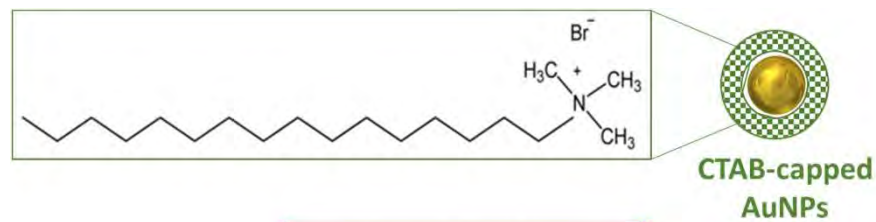
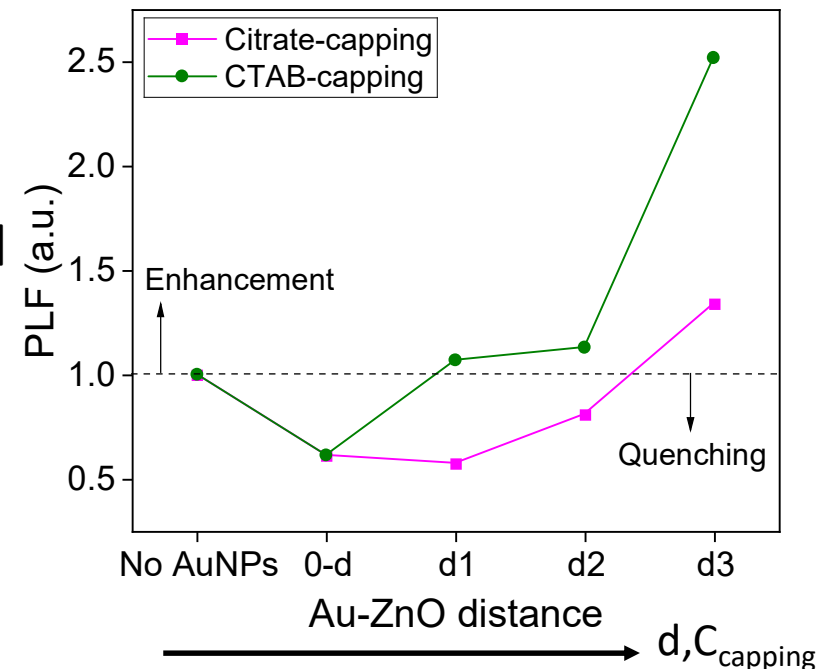
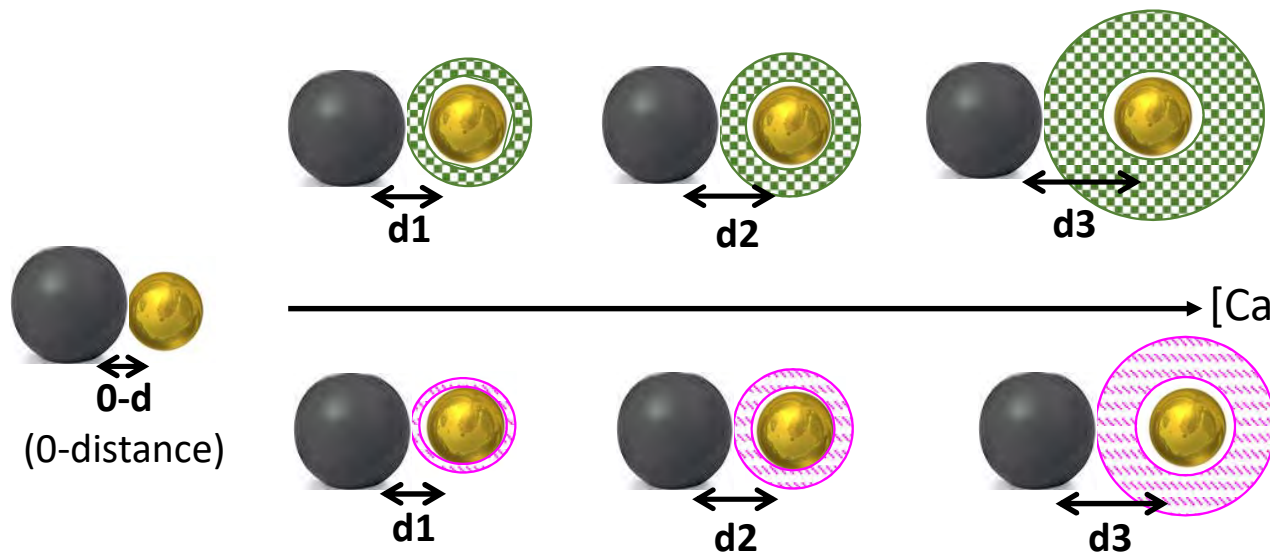
CTAB separated Au and ZnO



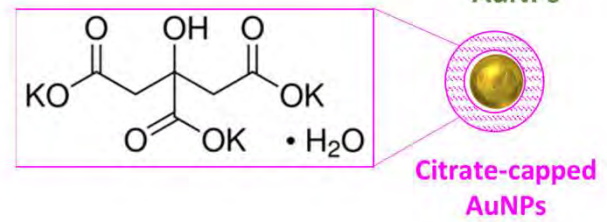
- PL enhancement → e⁻ transfer into ZnO
- CTAB capping of AuNPs
 - Creating distance between ZnO and AuNPs
 - Schottky barrier effect is reduced by the separation distance

What's the optimal heterojunction distance for photocatalysis

Effect of distance determined by capping agent concentration

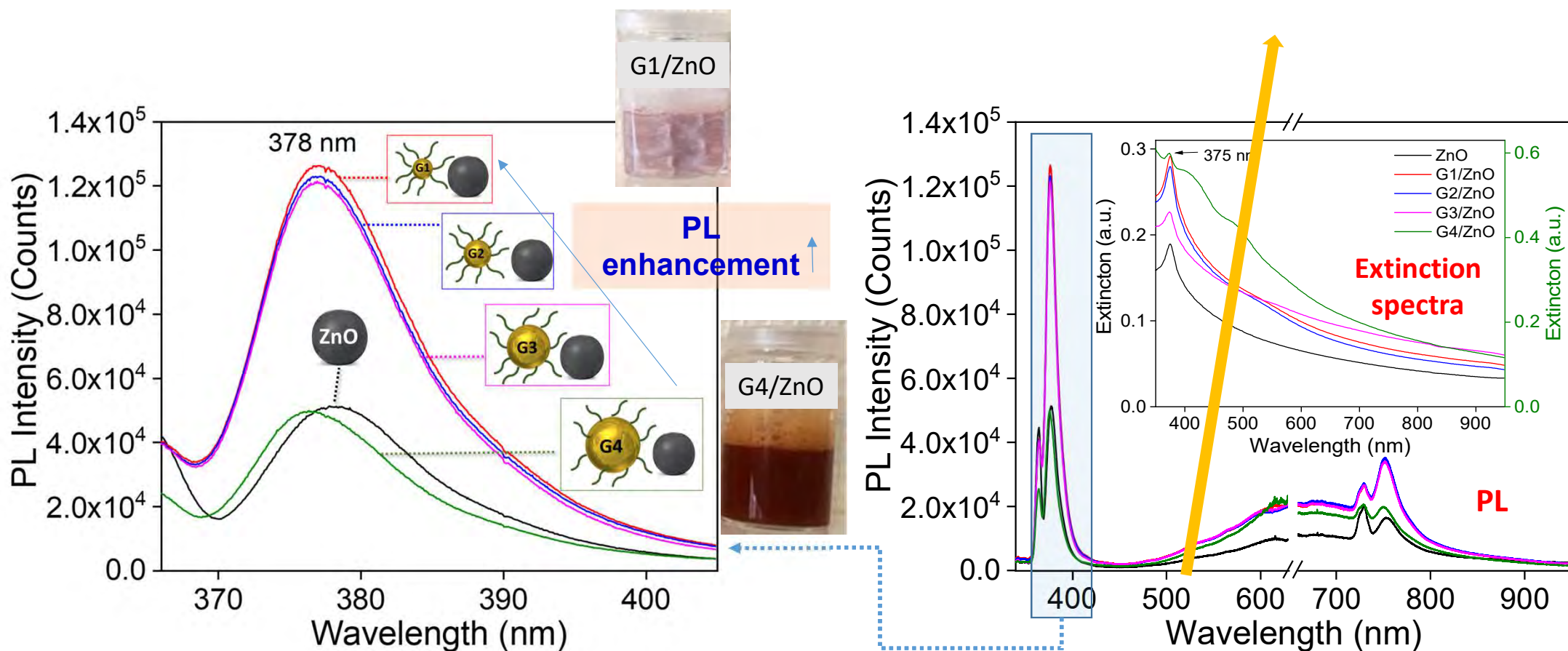


CTAB > Citrate
→ Distance by CTAB is larger



- Optimum interparticle distance
 - Inter-charge transfer between AuNPs ↓
 - Au-ZnO interactions ↑
- PL enhancement ↑

Effect of AuNPs size



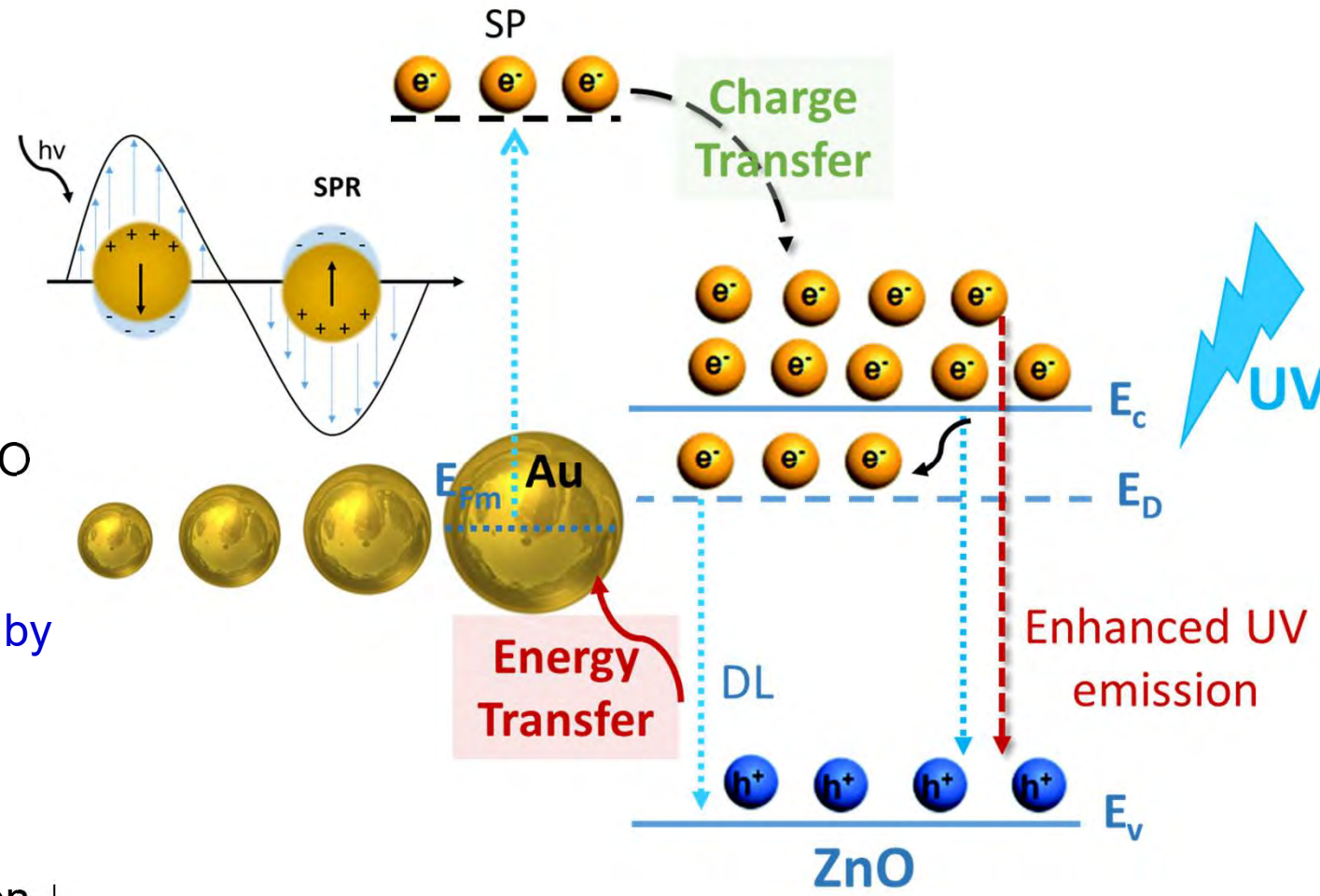
➤ PL enhancement under UV irradiation → **Indirect** charge transfer from AuNPs into ZnO NPs

Mechanism of PL enhancement

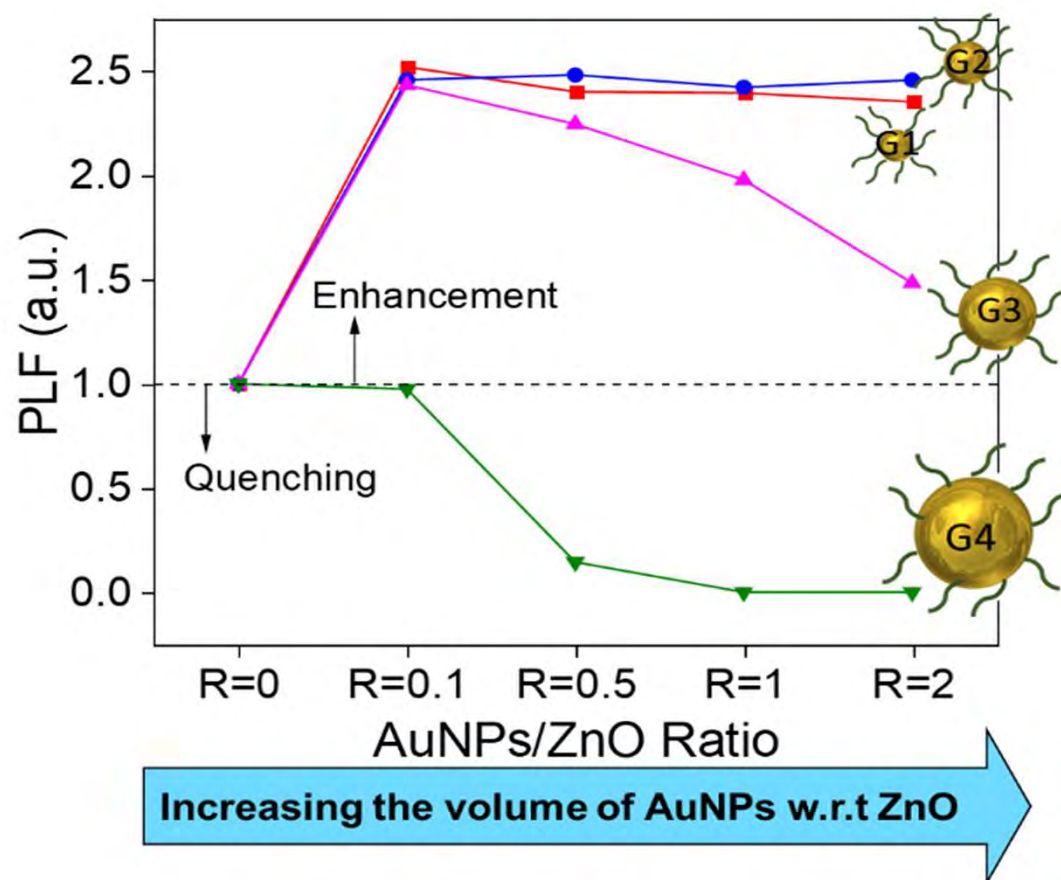
- Visible defect luminescence is absorbed by AuNPs (due to spectral overlap)
- Hot electrons in AuNPs are produced and transferred into ZnO

→ Indirect charge transfer induced by energy transfer

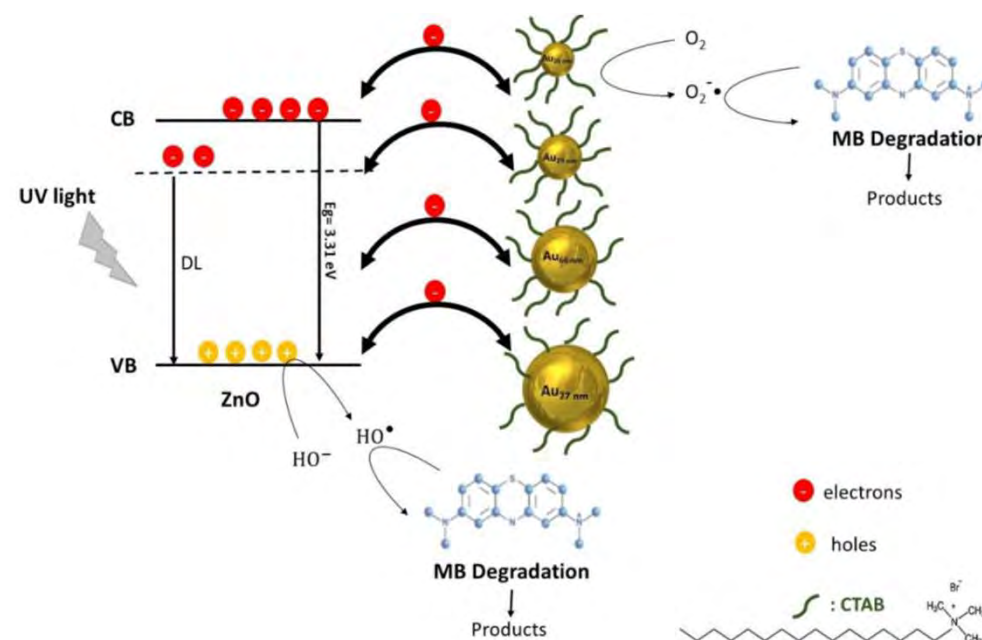
- As AuNPs size \uparrow
 - Au-ZnO interfacial interaction \downarrow
 - AuNPs Abs \downarrow , hot e^- \downarrow
- Less PL enhancement



Photocatalytic activity: Methylene blue degradation



Low quantities of small AuNPs:
Enhance the PL and PC
activities



Conclusions

The charge transfer (CT) based PL and PC is

- Defect mediated mechanism
- Plasmon structural features dependent
- Junction distance dependent

Enhancement: Optimum distances created by the gold's capping agent facilitated the indirect CT (AuNPs to ZnO) to enhance the PL by PIRET effect.

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